

Balancing Equations Answers

1. $2 \text{H}_2 + \text{O}_2 \rightarrow 2 \text{H}_2\text{O}$
2. $\text{H}_3\text{PO}_4 + 3 \text{KOH} \rightarrow \text{K}_3\text{PO}_4 + 3 \text{H}_2\text{O}$
3. $6 \text{K} + \text{B}_2\text{O}_3 \rightarrow 3 \text{K}_2\text{O} + 2 \text{B}$
4. $\text{HCl} + \text{NaOH} \rightarrow \text{NaCl} + \text{H}_2\text{O}$
5. $10 \text{Na} + 2 \text{NaNO}_3 \rightarrow 6 \text{Na}_2\text{O} + \text{N}_2$
6. $4 \text{C} + \text{S}_8 \rightarrow 4 \text{CS}_2$
7. $2 \text{Na} + \text{O}_2 \rightarrow \text{Na}_2\text{O}_2$
8. $2 \text{N}_2 + 5 \text{O}_2 \rightarrow 2 \text{N}_2\text{O}_5$
9. $2 \text{H}_3\text{PO}_4 + 3 \text{Mg(OH)}_2 \rightarrow \text{Mg}_3(\text{PO}_4)_2 + 6 \text{H}_2\text{O}$
10. $2 \text{NaOH} + \text{H}_2\text{CO}_3 \rightarrow \text{Na}_2\text{CO}_3 + 2 \text{H}_2\text{O}$
11. $\text{KOH} + \text{HBr} \rightarrow \text{KBr} + \text{H}_2\text{O}$
12. $\text{H}_2 + \text{O}_2 \rightarrow \text{H}_2\text{O}_2$
13. $4 \text{Na} + \text{O}_2 \rightarrow 2 \text{Na}_2\text{O}$
14. $2 \text{Al(OH)}_3 + 3 \text{H}_2\text{CO}_3 \rightarrow \text{Al}_2(\text{CO}_3)_3 + 6 \text{H}_2\text{O}$
15. $16 \text{Al} + 3 \text{S}_8 \rightarrow 8 \text{Al}_2\text{S}_3$
16. $6 \text{Cs} + \text{N}_2 \rightarrow 2 \text{Cs}_3\text{N}$
17. $\text{Mg} + \text{Cl}_2 \rightarrow \text{MgCl}_2$
18. $10 \text{Rb} + 2 \text{RbNO}_3 \rightarrow 6 \text{Rb}_2\text{O} + \text{N}_2$
19. $2 \text{C}_6\text{H}_6 + 15 \text{O}_2 \rightarrow 12 \text{CO}_2 + 6 \text{H}_2\text{O}$
20. $\text{N}_2 + 3 \text{H}_2 \rightarrow 2 \text{NH}_3$
21. $2 \text{C}_{10}\text{H}_{22} + 31 \text{O}_2 \rightarrow 20 \text{CO}_2 + 22 \text{H}_2\text{O}$
22. $\text{Al(OH)}_3 + 3 \text{HBr} \rightarrow \text{AlBr}_3 + 3 \text{H}_2\text{O}$
23. $2 \text{CH}_3\text{CH}_2\text{CH}_2\text{CH}_3 + 13 \text{O}_2 \rightarrow 8 \text{CO}_2 + 10 \text{H}_2\text{O}$
24. $\text{C} + \text{O}_2 \rightarrow \text{CO}_2$
25. $\text{C}_3\text{H}_8 + 5 \text{O}_2 \rightarrow 3 \text{CO}_2 + 4 \text{H}_2\text{O}$
26. $3 \text{Li} + \text{AlCl}_3 \rightarrow 3 \text{LiCl} + \text{Al}$
27. $2 \text{C}_2\text{H}_6 + 7 \text{O}_2 \rightarrow 4 \text{CO}_2 + 6 \text{H}_2\text{O}$
28. $3 \text{NH}_4\text{OH} + \text{H}_3\text{PO}_4 \rightarrow (\text{NH}_4)_3\text{PO}_4 + 3 \text{H}_2\text{O}$
29. $3 \text{Rb} + \text{P} \rightarrow \text{Rb}_3\text{P}$
30. $\text{CH}_4 + 2 \text{O}_2 \rightarrow \text{CO}_2 + 2 \text{H}_2\text{O}$
31. $2 \text{Al(OH)}_3 + 3 \text{H}_2\text{SO}_4 \rightarrow \text{Al}_2(\text{SO}_4)_3 + 6 \text{H}_2\text{O}$
32. $2 \text{Na} + \text{Cl}_2 \rightarrow 2 \text{NaCl}$
33. $16 \text{Rb} + \text{S}_8 \rightarrow 8 \text{Rb}_2\text{S}$
34. $2 \text{H}_3\text{PO}_4 + 3 \text{Ca(OH)}_2 \rightarrow \text{Ca}_3(\text{PO}_4)_2 + 6 \text{H}_2\text{O}$
35. $\text{NH}_3 + \text{HCl} \rightarrow \text{NH}_4\text{Cl}$
36. $2 \text{Li} + 2 \text{H}_2\text{O} \rightarrow 2 \text{LiOH} + \text{H}_2$
37. $\text{Ca}_3(\text{PO}_4)_2 + 3 \text{SiO}_2 + 5 \text{C} \rightarrow 3 \text{CaSiO}_3 + 5 \text{CO} + 2 \text{P}$
38. $4 \text{NH}_3 + 3 \text{O}_2 \rightarrow 2 \text{N}_2 + 6 \text{H}_2\text{O}$
39. $4 \text{FeS}_2 + 11 \text{O}_2 \rightarrow 2 \text{Fe}_2\text{O}_3 + 8 \text{SO}_2$
40. $5 \text{C} + 2 \text{SO}_2 \rightarrow \text{CS}_2 + 4 \text{CO}$
41. $\text{Fe} + \text{S} \rightarrow \text{FeS}$
42. $2 \text{KClO}_3 \rightarrow 2 \text{KCl} + 3 \text{O}_2$
43. $4 \text{Al} + 3 \text{O}_2 \rightarrow 2 \text{Al}_2\text{O}_3$
44. $\text{Fe}_2\text{O}_3 + 3 \text{C} \rightarrow 3 \text{CO} + 2 \text{Fe}$
45. $\text{K}_2\text{O} + \text{H}_2\text{O} \rightarrow 2 \text{KOH}$
46. $\text{K}_2\text{CO}_3 + \text{BaCl}_2 \rightarrow 2 \text{KCl} + \text{BaCO}_3$
47. $\text{Mg(OH)}_2 + \text{H}_2\text{SO}_4 \rightarrow \text{MgSO}_4 + 2 \text{H}_2\text{O}$
48. $2 \text{KF} + \text{BaBr}_2 \rightarrow \text{BaF}_2 + 2 \text{KBr}$
49. $\text{HCl} + \text{NH}_3 \rightarrow \text{NH}_4\text{Cl}$
50. $\text{Bi}_2(\text{SO}_4)_3 + 6 \text{NH}_4\text{OH} \rightarrow 2 \text{Bi(OH)}_3 + 3 (\text{NH}_4)_2\text{SO}_4$

Physical Science Balancing Answer Key

Judd E. Hollander



Physical Science Balancing Answer Key:

Physical Science Balancing Answer Key Book Review: Unveiling the Magic of Language

In an electronic era where connections and knowledge reign supreme, the enchanting power of language has been apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Physical Science Balancing Answer Key**," written by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound effect on our existence. Throughout this critique, we shall delve into the book's central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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