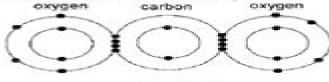


MERCY JUNIOR COLLEGE Grade 11 Physical Science Finals Exam

NAME:

YEAR LEVEL & SECTION: TEACHER'S NAME: PROCTOR'S NAME:

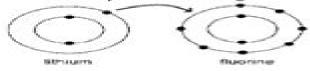
- I. MULTIPLE CHOICE INSTRUCTION: Encircle the letter of the correct answer. (1 POINT EACH)
- 1. It is a subatomic particle, symbol e- or
 - β-, with a negative elementary electric charge.
 - A. Neutron
 - 8. Electron
 - C. Protom
 - D. Attorni
- It is a subatomic particle, symbol p or p+, with a
 positive electric charge of +1e elementary
 charge and mass slightly less than that of a
 meutron.
 - A. Neutron
 - 8. Electron
 - C. Protom
 - D. Altern
- It is a subatomic particle, symbol n or nO, with no net electric charge and a mass slightly larger than that of a proton.
 - A. Neutron B. Electron
 - C. Proton
 - D. Atom
- 4. Is this an example of ionic bonding?



carbon dioxide

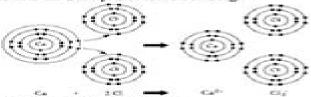
- A. Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- C. There is no chemical bonding present.
- D. This is just an orbital model.

5. Is this an example of ionic bonding?



littreum fluoride

- A. Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- There is no chemical bonding present.
- D. This is just an orbital model.
- In atomic physics and quantum chemistry, the electron configuration is the distribution of electrons of an atom or molecule in atomic or molecular orbitals.
 - A. Electron Configuration
 - Periodic Table of Elements
 - C. Orbital Models
 - D. Emergy Levels
- 7. Is this an example of ionic bonding?



- A. Yes, it is ionic bonding.
- No, it is of another type of chemical bonding.
- There is no chemical bonding present.
- This is just an orbital model.
- It is a chemical rule of thumb that reflects observation that atoms of main-group elements tend to combine in such a way that each atom has eight electrons in its valence shell, giving it the same electronic configuration as a noble gas.
 - A. Octet Bulle
 - 8. Valence Electrons
 - C. Atomic Orbitals
 - D. Chemical Bonds
- It is the smallest constituent unit of ordinary matter that has the properties of a chemical element.
 - A. Nieutron
 - B. Electron
 - C. Proton
 - D. Atom
- 10. It is a type of chemical bond that involves the electrostatic attraction between oppositely

Physical Science Grade 11 Final Examination 2014 Scope

M Mark

Physical Science Grade 11 Final Examination 2014 Scope:

This book delves into Physical Science Grade 11 Final Examination 2014 Scope. Physical Science Grade 11 Final Examination 2014 Scope is a vital topic that must be grasped by everyone, ranging from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Physical Science Grade 11 Final Examination 2014 Scope, encompassing both the fundamentals and more intricate discussions.

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 - Chapter 3: Physical Science Grade 11 Final Examination 2014 Scope in Everyday Life
 - Chapter 4: Physical Science Grade 11 Final Examination 2014 Scope in Specific Contexts
 - ∘ Chapter 5: Conclusion
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- 6. In chapter 5, the author will draw a conclusion about Physical Science Grade 11 Final Examination 2014 Scope. This chapter will summarize the key points that have been discussed throughout the book.
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