

Science

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(Chapter - 1) (Matter in our Surroundings) (Exercises)

(Class - IX)

Question 1:

Convert the following temperatures to the Celsius scale.

(a) 293 K

(b) 470 K

Answer 1:

(a) $293\text{ K} = (293 - 273)^\circ\text{C} = 20^\circ\text{C}$

(b) $470\text{ K} = (470 - 273)^\circ\text{C} = 197^\circ\text{C}$

Question 2:

Convert the following temperatures into the Kelvin scale.

(a) 25°C

(b) 373°C

Answer 2:

(a) $25^\circ\text{C} = (25 + 273)\text{ K} = 298\text{ K}$

(b) $373^\circ\text{C} = (373 + 273)\text{ K} = 646\text{ K}$

Question 3:

Give reasons for the following observations.

(a) Naphthalene balls disappear with time without leaving any solid.

(b) We can get the smell of perfume sitting several meters away.

Answer 3:

(a) Naphthalene shows the property of sublimation. Evaporation of naphthalene takes place easily and so it disappears during course of time without leaving a solid.

(b) Perfumes vaporize very fast and its vapours diffuse into air easily. That is why we can smell perfume sitting several meters away.

Question 4:

Arrange the following substances in increasing order of forces of attraction between the particles – water, sugar, oxygen.

Answer 4:

Oxygen < Water < Sugar.

Question 5:

What is the physical state of water at –

(a) 25°C

(b) 0°C

(c) 100°C ?

Answer 5:

(a) Liquid

(b) Solid and Liquid

(c) Liquid and Vapours

Question 6:

Give two reasons to justify:

(a) Water at room temperature is a liquid.

(b) An iron almirah is a solid at room temperature.

Answer 6:

(a) Water at room temperature is a liquid because it has fluidity and has definite volume but no definite shape.

(b) An iron almirah is a solid at room temperature because it is rigid and has a definite shape.

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