

What is a Report and an Essay in Academia

There are some basic differences between a report and an essay in an academic setting. The format of a report and essay differ as well as the main purpose of each. While there are similarities and differences between an essay and report, it's essential to know which format you're being asked to write. First, we'll discuss what a report is, and then continue on to see what makes an essay. Keep in mind that a report can be much longer than an essay, and report readers generally only quickly scan it to pick up its general idea. In either way, your ideas should be easily found and compelling enough so the reader continues to read.

To begin, a report and essay are similar in the fact that they both need to be in a formal style, have analytical thinking, a neat presentation with careful proofreading, as well as an introduction, body, and conclusion. Furthermore, a few distinct characteristics make a report different from an essay, such as a report gives information instead of an argument. Usually, a reader will more thoroughly read an essay compared to a report. Reports can include graphics and have short paragraphs along with numbered headings and sub-headings. Also, a writer will often need to write an executive summary after they've finished writing their full report. Such a summary isn't necessary with an essay.

Report

A report should be used to clearly and sufficiently inform the reader of the topic you're writing about. While every lecturer or professor may have different guidelines they prefer, there are general rules to follow when writing a report. Unless otherwise requested, a report should consist of the following essential sections: title page, introduction, body, and conclusion. If you want to go a step further, include a letter of transmittal, table of contents, list of abbreviations and/or a glossary, executive summary, recommendations, bibliography, and appendices.

The style and presentation of a report are very important in order to be taken seriously and have people want to read your full report. There are certain tips that you should follow to write a report that leaves a great first impression. You'll want to ensure you use plenty of white space, and that the different parts of your report are easily recognized. Don't forget to use sub-headings with plenty of space between different sections. Graphs, illustrations, and maps are encouraged to be used as they can clarify the information you're trying to give. Number each page of a report with consistent formatting, and always use formal language.

On the other hand, there are some things to avoid when writing a report. Leave out any inaccurate and conflicting information. Don't include any outdated data, or irrelevant statistics. Keep opinions separate from facts, and leave out any unsupported recommendations or conclusions. Also, don't let somebody read your report if it focuses more on appearance and has a lack of content. Lastly, a report isn't finished without careful proofreading.

The topics for reports usually consist of a problem or case study with a hypothetical situation. The information for reports comes from readings, fieldwork, and practical work. A report's purpose is not

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L Darling-Hammond



Sample Of Report Essay:

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the objectives of the boiling point determination lab were to determine the boiling point of a mystery substance provided to each student boiling point is the point at which a liquid changes physical states and becomes a vapor or gas boiling point is important to understand because it can be used to help distinguish between different substances

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the boiling point of an organic liquid is the temperature at which its vapour pressure equals the atmospheric pressure over the liquid or it is the temperature at which the vapour and liquid phases are in equilibrium at agiven pressure the boiling point is considered as a criterion of purity of a compound and is useful for identification of

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given that the boiling points of ethanol and methanol 78 c and 64 c respectively are far off in a significant way this seems like a valid conclusion conclusion thermometer calibrations are necessary in order to collect reliable information as precise calculations are imperative to gathering correct results

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melting point 1 a push the open end of a capillary tube into the powdered acetamide b move the powder to the closed end of the capillary tube by tapping it on the table repeat until the the powdered acetamide occupies 1 2 mm of the capillary tube end c

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boiling point determination lab report conclusion objectivesto correctly measure boiling issues of various liquids and identify an unknown liquid based on its boiling point background information objekte can be distinguished into acid properties and physical properties

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discussion and conclusion the technique recrystallization was used because it is a simple method to purify a solid compound the percentage yield for benzoic acid and acetanilide acid being high indicated the initial sample had a low concentration of impurities with the melting point being in a specific

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introduction in this lab we will be using boiling points melting points and infrared spectrometry to identify unknown compounds boiling points are affect by three key factors the first being intermolecular forces inter molecular affect boiling point based on bond strength

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determination of boiling points introduction the boiling point of a compound is the temperature at which it changes from a liquid to a gas this is a physical property often used to identify substances or to check the purity of the compound it

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clean up solid and liquid spills immediately true g conclusions after conducting the boiling point determination lab it can be concluded that the unknown substance 193 is isopropyl alcohol on account of its observed boiling point of 88 c

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conclusion after completing this experiment the melting point of three different unknown compounds was obtained using the mel temp apparatus compound a had a melting point range of 114 6 115 5 c thus revealing it to be acetanilide compound b had a melting point range of 122 6 124 5 c thus

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nov 27 2019 essay sample the purpose of this experiment is to determine the boiling points of various organic compounds and to use these to identify unknowns equipment free essays

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melting point can be used as the identity and purity of an organic solid however it is not the same as the boiling point the boiling point is a characteristic unique of pure liquids the boiling point is the temperature at which the vapor pressure of a liquid is equal to the atmospheric pressure

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aug 11 2020 boiling points are very sensitive to changes in applied pressure so all boiling points should be reported with the measured pressure a compound s normal boiling point refers to its boiling point at a pressure of 760 text mm ce hg

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the purpose of the experiment is to understand what vapour pressure is and how it changes with temperature to understand the relationship between vapour pressure at a given temperature and boiling temperature at a given pressure to determine

the vapour pressure of a pure liquid at various temperatures

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exploring boiling points abstract in this experiment we are given a set of straight chain alkanes and a set of compounds to compare by following the trendline of the straight chain alkanes it could be concluded that as the molar mass of the straight chain alkanes increases their boiling points increase as well

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laboratory partners samantha and mackenzie goals □ correctly determine the boiling point of an unknown liquid □ measure the refractive index of an unknown □ use the gas chromatogram to determine the mass percentage of a mixture and

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50 °C so in order to convert an experimental measurement taken in Calgary which is at higher altitude than sea level and so is at a lower pressure to that reported for sea level higher pressure one needs to add a correction factor since the boiling point at sea level is higher than that at higher altitudes

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introduction in chemistry one may encounter an unknown substance that needs to be determined by physical properties such as boiling point melting point density solubility etc in this experiment I aim to figure out the boiling point of a mystery substance in order to identify it chemical responsibility this experiment requires the handling of a Bunsen burner boiling

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there are a variety of methods by which a sample's boiling point can be determined including distillation reflux and by using a Thiele tube the most straightforward method uses a Thiele tube and has the advantage of using less than 0.5 mL of material

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