

FUNCTIONAL GROUPS OF ORGANIC COMPOUNDS

Homologous series	Structure of functional group	Example of a compound	
		Name	Structural formula
alkanes	Only C-H and C-C single bonds	ethane	
alkenes		ethene	
alkynes		ethyne	
haloalkanes (alkyl halides)	 (X = F, Cl, Br, I)	bromoethane	
alcohols (alkanols)		ethanol	
aldehydes		ethanal	
ketones		propan-2-one	
carboxylic acids		ethanoic acid	
esters		methyl ethanoate	

NAMING OF ORGANIC COMPOUNDS

The name of every organic molecule has three parts:

- The parent name indicates the number of C atoms in the longest carbon chain in the molecule.
- The suffix indicates what functional group is present.
- The prefix reveals the identity, location and number of substituents attached to the carbon chain.

prefix

What and where are the substituents?

parent

How many carbons?

suffix

What is the functional group/homologous series?

di - 2

tri - 3

tetra - 4

Number of carbon atoms	Parent name	Number of carbon atoms	Parent name
1	meth	5	pent
2	eth	6	hex
3	prop	7	hept
4	but	8	oct



prefix, suffix and parent!

Page 4 of 41
Tetra - chloro - metha

Physical Science Organic Compounds Study Guide

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A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the EduGorilla Prep Experts text.

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Table of Contents Physical Science Organic Compounds Study Guide

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 - The Rise of Digital Reading Physical Science Organic Compounds Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Organic Compounds Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Organic Compounds Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Organic Compounds Study Guide
 - Personalized Recommendations
 - Physical Science Organic Compounds Study Guide User Reviews and Ratings
 - Physical Science Organic Compounds Study Guide and Bestseller Lists
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 - Highlighting and Note-Taking Physical Science Organic Compounds Study Guide
 - Interactive Elements Physical Science Organic Compounds Study Guide
- 8. Staying Engaged with Physical Science Organic Compounds Study Guide
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 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Organic Compounds Study Guide
- 9. Balancing eBooks and Physical Books Physical Science Organic Compounds Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Organic Compounds Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Organic Compounds Study Guide
 - Setting Reading Goals Physical Science Organic Compounds Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Organic Compounds Study Guide
 - Fact-Checking eBook Content of Physical Science Organic Compounds Study Guide
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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