

Periodic Table Basics

Step 1: Complete the square for each element by filling in the atomic number, name, & atomic mass.

Step 2: Determine the number of protons, neutrons, and electrons in an atom of each element.

Step 3: Identify if the element is a solid, liquid, or gas at room temperature.

Step 4: Give the melting (M.P.) and boiling points (B.P.) in degrees Celsius.

Step 5: List at least three physical or chemical properties for each element.

Step 6: List at least three uses for each element.

Step 7: Draw a Bohr diagram and Lewis Structure to show the arrangement of electrons and the number of valence electrons.

Step 8: Use the following colors to shade in the square for each element. You should **ONLY** color the small square in the upper left-hand corner and not the entire card.

Green = Li & Na
Orange = B & Al

Pink = O & S
Red = C & Si

Blue = Be & Mg
Tan = N & P

Purple = F & Cl
Yellow = He, Ne, & Ar

Step 9: Cut the cards apart and arrange according to atomic number in the pattern shown. Once you have the cards arranged in the correct order, glue them to a large sheet of construction paper.

1							2
3	4	5	6	7	8	9	10
11	12	13	14	15	16	17	18

① B 	P= ② N= E=	③ O S O L O G
	M.P. = B.P. = ④	
Properties ⑤		
Uses ⑥		
Bohr Diagram Lewis Structure		
 ⑦ B		

Done? Answer the questions on the worksheet using the information on your Periodic Table!

Physical Science If8767 Answers Percentage Composition

Rachel S Tattersall



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Chemical Calculations Richard Lloyd Whiteley, 1908 Percent Composition and Empirical Formulas, 2016 In this lecture you encounter two classic types of chemistry problems and learn the basic characteristics of each The lecture concludes with several practice problems to help you master the skill of solving percent composition problems

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