

Chemical Formula

Talking about proportions, molecules and relative of atoms indicated by a subscript.

Molecular $C_{10}H_{22}$ (molecular)

Formula represents the number of _____ of _____
in a single molecule.

Empirical C_5H_{11} (molecular)

Formula represents the simplest ratio of
positive and negative ions in a formula
unit.

Atomic Mass

The average mass of atoms relative to hydrogen
atoms together.

Isotopes

$^{12}_6C$ stable isotope carbon

$^{13}_6C$ stable isotope carbon

Radioisotopes

$^{14}_6C$ radioisotope carbon

$^{15}_6C$ radioisotope carbon

$^{16}_6C$ radioisotope

Note Taking Guide Episode 601 Chemical Formula

Rachel Sandford



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