

Chemical Formula

Talking about proportions, molecules and relative of atoms indicated by a subscript.

Molecular $\text{C}_{10}\text{H}_{22}$ (molecules)

Formula represents the number of _____ of _____
in a single molecule.

Atomic C (atoms) (molecules)

Formula represents the proportions of _____ of
positive and negative _____ atoms formula
unit.

Atomic Mass

The proportions of the atoms proportions of _____
atoms together.

Mass

$\text{C}_{10}\text{H}_{22}$ _____ atoms

_____ proportions of _____
atoms

Subscript

$\text{C}_{10}\text{H}_{22}$ _____ atoms

_____ proportions of _____

$\text{C}_{10}\text{H}_{22}$ _____

Note Taking Guide Physics 601

Marco Cascella



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