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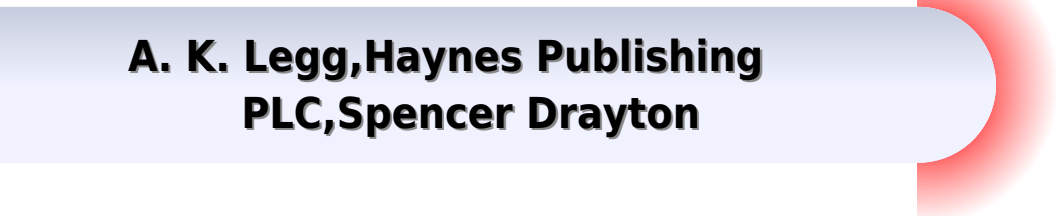
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Samsung Prevail Repair Manual

**A. K. Legg, Haynes Publishing
PLC, Spencer Drayton**



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element are isotopes of each other ordinary beans are a lot bigger than atoms but perhaps they can give you one or two clues about isotopes we have talked in class about isotopes atoms of the same element that differ in

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web beanium isotope b brownium beanium isotope c whiteium of atoms beans mass beans in cup mass empty cup mass of atoms beans 2 minus 3 average mass of 1 bean 4 divided by 1 of beans 1 a b c x 100 section 3 calculate the average atomic mass of beanium

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web in the beanium lab students obtained a sample of beanium a fictitious element in order to learn about isotopes and average atomic mass beanium has three isotopes white brown and speckled which have different masses and different percent abundances

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