

34	a) How $^{35}\text{Cl}_{17}$ and $^{37}\text{Cl}_{17}$ are related to each other? b) If both of these forms of Cl occurs in nature as 75% and 25% respectively. Calculate the average atomic mass of Chlorine atom. c) Draw the labelled schematic atomic structure of Chlorine atom having $Z=17$. Or a) Distinguish between the properties of electrons, protons, and neutrons with respect to position, charge and mass. b) Write the valency and atomic symbol of the following element: i. Sodium ($Z=11$) ii. Aluminium ($Z=13$) iii. Potassium ($Z=19$) iv. Calcium ($Z=20$)	5
35	a) Plastids are present only in the plant. Which types of plastids are present in : i. Petals; ii. Leaves; iii. fruit pulp b) What happen if chloroplast is removed from the plant. Explain. Or To test the effect of the concentration of the solution on the cell a student placed some plant cell and animal cell (RBC) for 15 minute in the hypertonic solution and then shifted them into hypotonic solution. Answer the following question: a) What is hypotonic solution. b) What will the expected observation in the above situation. c) what is the role of cell wall in the cell in above situation.	5
36	a) Explain universal law of gravitation also write its mathematical expression. b) Write three importance of this law. Or Give reason: a) carry a bag with broader strap is easy than a thin strap bag. b) why sharp knife cut easy in comparison with the blunt knife. c) camel can run in a desert easily.	5
SECTION-E Q.no.37 to 39 are case based / source- based questions with 2 to 3 short sub - parts.		

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