

# CHEMISTRY

2013

Time: 30 minutes

Max. Marks: 15

## SECTION "A" (MULTIPLE CHOICE QUESTIONS)

1. Choose the correct answer for each from the given options:

- (i) it is a trial idea:  
(a) Experiment (b) Hypothesis (c) Theory (d) Scientific Law
- (ii) It is the empirical formula of glucose:  
(a) CH<sub>2</sub>O (b) C<sub>2</sub>H<sub>4</sub>O<sub>2</sub> (c) C<sub>5</sub>H<sub>10</sub>O<sub>5</sub> (d) C<sub>6</sub>H<sub>12</sub>O<sub>6</sub>
- (iii) It is not an isotope of oxygen:  
(a) <sup>16</sup><sub>8</sub>O (b) <sup>17</sup><sub>8</sub>O (c) <sup>18</sup><sub>8</sub>O (d) <sup>19</sup><sub>8</sub>O
- (iv) The vertical columns of the periodic table are called:  
(a) Periods (b) Series (c) Groups (d) Rows
- (v) It is an ionic compound:  
(a) H<sub>2</sub>O (b) CO<sub>2</sub> (c) NH<sub>3</sub> (d) NaCl
- (vi) Neither definite shape nor definite volume is the property of:  
(a) Solid (b) Liquid (c) Gas (d) None of them
- (vii) It is defined as the number of moles of solution dissolved in one litre of solution:  
(a) Molarity (b) Molality (c) Normality (d) Mole fraction
- (viii) It is the unit of electrical current:  
(a) volt (b) coulomb (c) faraday (d) ampere
- (ix) It is a proton-donor:  
(a) Acid (b) Base (c) Salt (d) Alkali
- (x) It is an exothermic process:  
(a) Evaporation (b) Respiration  
(c) Thermal decomposition (d) Melting
- (xi) It is an ion-exchanger resin:  
(a) Na-zeolite (b) Ca(OH)<sub>2</sub> (c) MgSO<sub>4</sub> (d) NaCl
- (xii) When carbon burns in air, this gas is produced:  
(a) SO<sub>2</sub> (b) H<sub>2</sub>S (c) NH<sub>3</sub> (d) CO<sub>2</sub>
- (xiii) It is the study of Hydrocarbons and their derivatives:  
(a) Physical Chemistry (b) Inorganic Chemistry  
(c) Organic Chemistry (d) Environmental Chemistry
- (xiv) This gas is used for respiration:  
(a) Hydrogen (b) Oxygen (c) Nitrogen (d) Chlorine
- (xv) It is the only liquid metal:  
(a) Bromine (b) Mercury (c) Gold (d) Sulphur
- (xvi) The most abundant and useful halogen is:  
(a) Fluorine (b) Chlorine (c) Bromine (d) Iodine
- (xvii) Red ink is normally prepared from:

# Pixl Past Papers Chemistry November 2014

**DJ Losen**



**Pixl Past Papers Chemistry November 2014:**

The book delves into Pixl Past Papers Chemistry November 2014. Pixl Past Papers Chemistry November 2014 is a vital topic that must be grasped by everyone, from students and scholars to the general public. This book will furnish comprehensive and in-depth insights into Pixl Past Papers Chemistry November 2014, encompassing both the fundamentals and more intricate discussions.

1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Pixl Past Papers Chemistry November 2014
- Chapter 2: Essential Elements of Pixl Past Papers Chemistry November 2014
- Chapter 3: Pixl Past Papers Chemistry November 2014 in Everyday Life
- Chapter 4: Pixl Past Papers Chemistry November 2014 in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, this book will provide an overview of Pixl Past Papers Chemistry November 2014. The first chapter will explore what Pixl Past Papers Chemistry November 2014 is, why Pixl Past Papers Chemistry November 2014 is vital, and how to effectively learn about Pixl Past Papers Chemistry November 2014.

3. In chapter 2, this book will delve into the foundational concepts of Pixl Past Papers Chemistry November 2014. The second chapter will elucidate the essential principles that need to be understood to grasp Pixl Past Papers Chemistry November 2014 in its entirety.

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6. In chapter 5, this book will draw a conclusion about Pixl Past Papers Chemistry November 2014. The final chapter will summarize the key points that have been discussed throughout the book.

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## **Table of Contents Pixl Past Papers Chemistry November 2014**

1. Understanding the eBook Pixl Past Papers Chemistry November 2014
  - The Rise of Digital Reading Pixl Past Papers Chemistry November 2014
  - Advantages of eBooks Over Traditional Books
2. Identifying Pixl Past Papers Chemistry November 2014
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Pixl Past Papers Chemistry November 2014
  - User-Friendly Interface
4. Exploring eBook Recommendations from Pixl Past Papers Chemistry November 2014
  - Personalized Recommendations
  - Pixl Past Papers Chemistry November 2014 User Reviews and Ratings
  - Pixl Past Papers Chemistry November 2014 and Bestseller Lists
5. Accessing Pixl Past Papers Chemistry November 2014 Free and Paid eBooks
  - Pixl Past Papers Chemistry November 2014 Public Domain eBooks
  - Pixl Past Papers Chemistry November 2014 eBook Subscription Services
  - Pixl Past Papers Chemistry November 2014 Budget-Friendly Options
6. Navigating Pixl Past Papers Chemistry November 2014 eBook Formats
  - ePub, PDF, MOBI, and More
  - Pixl Past Papers Chemistry November 2014 Compatibility with Devices
  - Pixl Past Papers Chemistry November 2014 Enhanced eBook Features
7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Pixl Past Papers Chemistry November 2014
  - Highlighting and Note-Taking Pixl Past Papers Chemistry November 2014
  - Interactive Elements Pixl Past Papers Chemistry November 2014

8. Staying Engaged with Pixl Past Papers Chemistry November 2014
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Pixl Past Papers Chemistry November 2014
9. Balancing eBooks and Physical Books Pixl Past Papers Chemistry November 2014
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Pixl Past Papers Chemistry November 2014
10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
11. Cultivating a Reading Routine Pixl Past Papers Chemistry November 2014
  - Setting Reading Goals Pixl Past Papers Chemistry November 2014
  - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pixl Past Papers Chemistry November 2014
  - Fact-Checking eBook Content of Pixl Past Papers Chemistry November 2014
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
  - Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
  - Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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