education
MPUMALANGA PROVINCE
REPUBLIC OF SOUTH AFRICA

PHYSICAL SCIENCES (CHEMISTRY)
GRADE 11 TERM 3 FORMAL EXPERIMENT - PART 2 **TIME: 1 HOUR**

VERIFICATION OF BOYLE'S LAW
INDIVIDUAL WORK (should be done under strict supervision).

1. **Write a scientific report which must include the following:**
1.1 **Aim** (1)
1.2 **Apparatus and method** (5)

2. **Observations and Recording of results** (Copy the table below onto your book or sheet of paper and transfer the results you recorded when conducting the experiment). Calculate and pV for each set of results and record your answers in the table.

Volume (V) cm^3	Pressure (p) kPa		pV ( )

(6)

3. **Analysis of the results:**
3.1 Identify the following variables for the experiment:
- Independent
- Dependent
- Controlled (3)

3.2 Use the graph paper provided by your teacher to draw graphs of V versus p and V versus . (18)

4. **Conclusion**
4.1 Examine the results in your table as well as the two graphs you have drawn, what happens to the volume of the gas in the tube as the pressure is decreased. (1)
4.2 Write down a conclusion for this experiment. (3)

5
Questions and discussion:

5.1 Explain why you have to wait for a few minutes before you can take readings. (2)
5.2 Explain the relationship between pressure and volume in terms of the kinetic theory of gases. (4)

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David G. Myers

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Table of Contents Physical Science Paper Exampler Grade 11

1. Understanding the eBook Physical Science Paper Exampler Grade 11
    - The Rise of Digital Reading Physical Science Paper Exampler Grade 11
    - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Paper Exampler Grade 11
    - Exploring Different Genres
    - Considering Fiction vs. Non-Fiction
    - Determining Your Reading Goals
3. Choosing the Right eBook Platform
    - Popular eBook Platforms
    - Features to Look for in an Physical Science Paper Exampler Grade 11
    - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Paper Exampler Grade 11
    - Personalized Recommendations
    - Physical Science Paper Exampler Grade 11 User Reviews and Ratings
    - Physical Science Paper Exampler Grade 11 and Bestseller Lists

5. Accessing Physical Science Paper Exampler Grade 11 Free and Paid eBooks
    - Physical Science Paper Exampler Grade 11 Public Domain eBooks
    - Physical Science Paper Exampler Grade 11 eBook Subscription Services
    - Physical Science Paper Exampler Grade 11 Budget-Friendly Options
6. Navigating Physical Science Paper Exampler Grade 11 eBook Formats
    - ePub, PDF, MOBI, and More
    - Physical Science Paper Exampler Grade 11 Compatibility with Devices
    - Physical Science Paper Exampler Grade 11 Enhanced eBook Features
7. Enhancing Your Reading Experience
    - Adjustable Fonts and Text Sizes of Physical Science Paper Exampler Grade 11
    - Highlighting and Note-Taking Physical Science Paper Exampler Grade 11
    - Interactive Elements Physical Science Paper Exampler Grade 11
8. Staying Engaged with Physical Science Paper Exampler Grade 11
    - Joining Online Reading Communities
    - Participating in Virtual Book Clubs
    - Following Authors and Publishers Physical Science Paper Exampler Grade 11
9. Balancing eBooks and Physical Books Physical Science Paper Exampler Grade 11
    - Benefits of a Digital Library
    - Creating a Diverse Reading Collection Physical Science Paper Exampler Grade 11
10. Overcoming Reading Challenges
    - Dealing with Digital Eye Strain
    - Minimizing Distractions
    - Managing Screen Time
11. Cultivating a Reading Routine Physical Science Paper Exampler Grade 11
    - Setting Reading Goals Physical Science Paper Exampler Grade 11
    - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Physical Science Paper Exampler Grade 11
    - Fact-Checking eBook Content of Physical Science Paper Exampler Grade 11
    - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends
    - Integration of Multimedia Elements
    - Interactive and Gamified eBooks

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