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- Ⓐ Which one of the following decreases in a group from top to bottom?

51. Metallic nature

D) Electronegativity

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- Diagram illustrating the molecular orbital configuration for the N_2^+ ion. The diagram shows the energy levels for the N_2 molecule (left) and the N_2^+ ion (right). The N_2 molecule has a bond order of 3, while the N_2^+ ion has a bond order of 2.5. The diagram shows the removal of an electron from the antibonding π^* orbital of N_2 to form N_2^+ .

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3 - 2 = 6 M

10. Pose any two questions to understand the difference between valency and valency electrons.

11. Explain why bonding angle (HOH) in water is $104^{\circ} 31'$ instead of $109^{\circ} 28'$.

Physical Science Grade 1paper 23

Yicheng Fang



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