

Atomic Number

- Increase across a period
- increase down a family

Atomic Mass

- increase across a period
- increase down a family

Atomic Radius

half the distance between nuclei of two like atoms.

- decreases across a period (due to increased attraction of more positive nuclei to electron clouds)
- increase down a family (due to addition of an energy level)

Electron Affinity

amount of energy when an atom gains an electron

- increases across a period
- decreases down a family

Ionization Energy

amount of energy required to remove an electron from a neutral atom

- increase across a period
- decrease down a family

Electron Distribution - Forming Ions

- Atoms become ions by gaining or losing electrons to become an ion.
- An atom is stable when it has a full outer energy level (8 valence electrons).
- Atoms that lose (give away) electrons become positive charged ions. Atoms that gain (steal) electrons become negative charged ions.

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Readers' Guide to Periodical Literature Anna Lorraine Guthrie, Bertha Tannehill, Neltje Marie Tannehill Shimer, 1919
Pitman's Journal of Commercial Education, 1907 Readers' Guide to Periodical Literature, 1919 **Resources in Education**, 1997 The Routledge Handbook of Conference Interpreting Michaela Albl-Mikasa, Elisabet Tiselius, 2021-11-29
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