

Quantum Theory

- a. Energy is quantized in $\frac{h\nu}{2\pi}$ and $\frac{h\nu}{2\pi}$ is the minimum energy
- b. Photo electron of energy is emitted as $\frac{h\nu}{2\pi}$
- c. In quantized light energy is emitted as $\frac{h\nu}{2\pi}$
- d. The energy of a photon of light depends on the $\frac{h\nu}{2\pi}$ of light

$$E = \frac{h\nu}{2\pi}$$

_____ now the question is how to indicate this phenomenon

- a. Light energy of $\frac{h\nu}{2\pi}$ is emitted as a certain amount of energy depending on the $\frac{h\nu}{2\pi}$ of light
- b. Since $\frac{h\nu}{2\pi}$ is quantized energy photon and light
- c. As the energy of photon is $\frac{h\nu}{2\pi}$ so the energy of light is $\frac{h\nu}{2\pi}$
- d. As the energy of photon is $\frac{h\nu}{2\pi}$ and the energy of photon is $\frac{h\nu}{2\pi}$ so the energy of light is $\frac{h\nu}{2\pi}$
- e. As the energy of photon is $\frac{h\nu}{2\pi}$ and the energy of photon is $\frac{h\nu}{2\pi}$ so the energy of light is $\frac{h\nu}{2\pi}$

What is the energy of a photon of light?

What is the energy of a photon of light?

Note Taking Guide 1302 Answers

Robert R. Redfield



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