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Lingjun Ying



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type of chemical reaction for this reaction we have a neutralization reaction balancing strategies here we have a neutralization reaction

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web the balanced chemical equation for this reaction is $\text{mg s} + 2\text{hno}_3 \text{ aq} \rightarrow \text{mg no}_3 \text{ 2 aq} + \text{h}_2 \text{ g}$ in this equation the s symbolizes solid magnesium aq represents aqueous nitric acid aq signifies magnesium nitrate in a dissolved state and g represents the gaseous hydrogen produced

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web mar 4 2018 $\text{mg} + 2\text{hno}_3 \rightarrow \text{mg no}_3 \text{ 2} + \text{h}_2$ on the left side there are mg 1 magnesium in elemental form h 2 there are 2 molecules of nitric acid each acid has one hydrogen so two molecules of nitric acid is 2 hydrogens n 2 there are 2 molecules of nitric acid each acid has one nitrogen so two molecules of nitric acid is 2 nitrogens o 6

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and two moles of nitric acid HNO_3 react to form one mole of magnesium nitrate $\text{Mg}(\text{NO}_3)_2$ one mole of water H_2O and one mole of carbon dioxide CO_2 show chemical structure image reactants

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balanced total equation $\text{H}_2\text{SO}_4 + 2\text{LiOH} \rightarrow \text{Li}_2\text{SO}_4 + 2\text{H}_2\text{O}$