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Acids, Bases, and Solutions

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Across

- 2. The strongest bases release this ion in water
- 4. Describes a solution where a lot of solute is dissolved in the water
- & Solutes lower the ______ point of a
- Solution remains this way if you can continue to dissolve more solute
- Solution becomes this when so much solute is added that no more dissolves.
- 10. A solution where only a little solute is dissolved in the water
- 53. In a weell-mixed mixture
- 14. Is a mirture in which perticles can be seen and easily separated by setting or filtration
- 16. A compound that changes color when in contact with an acid or a base
- till. Most acids release this ion in water
- 18. Any lonic compound made from a neutralization of an add with a base

Dicharts

- 1. This type of acid-base reaction or atuces water and a sait
- Is a substance that testes sour, reacts with metals, and turns blue litrus paper red
- Property of acids that describes their ability to "est away" at other materials.
- 6. One factor that affects solubility [Hint: can be measured in degrees Celelus]
- is a measure of how well a solute can dissolve in a solvent at a given immperature
- 11. Solutes raise the _____ point of a solvent
- 12. The type of ______ also affects solubility, whether it's polar or non-polar
- The part of a solution that is present in a solution in a arrailer amount and dissolved by the solvent
- is a substance that testes bitter, feels slippery, and turns red itmus paper blue.

Physical Science Grade Chapter Acids Bases And Solutions Crossword Key

Jianjun Gao

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