

PERIODIC TABLE (BASIC)

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Hybrid program

Hydrogen is a chemical element with chemical symbol H and atomic number 1. With an atomic weight of 1.00794 u, hydrogen is the lightest element on the periodic table. Its monatomic form (H) is the most abundant chemical substance in the universe, constituting roughly 75% of all baryonic mass. Non-remnant stars are mainly composed of hydrogen in its plasma state. The most common isotope of hydrogen, termed protium (name rarely used, symbol ${}^1\text{H}$), has a single proton and zero neutrons.

High Intensity

Helium is a chemical element with symbol He and atomic number 2. It is a colorless, odorless, tasteless, non-toxic, inert, monoatomic gas that heads the noble gas group in the periodic table. Its boiling and melting points are the lowest among the elements and it exists only as a gas except in extreme conditions.

Unit 1

Lithium (from Greek: λίθος lithos, "stone") is a chemical element with symbol Li and atomic number 3. It is a soft, silver-white metal belonging to the alkali metal group of chemical elements. Under standard conditions it is the lightest metal and the least dense solid element. Like all alkali metals, lithium is highly reactive and flammable. For this reason, it is typically stored in mineral oil. When cut open, lithium exhibits a metallic luster, but contact with moist air corrodes the surface quickly to a dull silvery gray, then black tarnish. Because of its high reactivity, lithium never occurs freely in nature, and instead, only appears in compounds, which are usually ionic. Lithium occurs in a number of pegmatitic minerals, but due to its solubility as an ion it is present in ocean water and is commonly obtained from brines and clays.

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Clemens Wendtner

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Guides) Speedy Publishing LLC, 2014-09-02 If a student does not have a strong background chemistry and biology or if their knowledge is a little rusty and could use some refreshing a fundamentals guide to organic chemistry is an excellent way to get back into the swing of things A guide will go over the different types of bonds atomic structure and cover the important topics associated with biology that the student will need to understand before entering an organic chemistry class Furthermore study guides are much cheaper than a tutor taking another course or a textbook The guide is succinct too and will not cover unnecessary or ancillary topics that will not be relevant to an organic chemistry course

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