

**Periodic Trends**

1. Explain the following trends:

- Atomic Radius
- Electronegativity
- Ionization Energy

2. Order the following elements by increasing atomic radius:

Li, Na, K, Rb, Cs, Fr

3. Order the following elements by increasing electronegativity:

F, Cl, Br, I, At

4. Order the following elements by increasing ionization energy:

Li, Na, K, Rb, Cs, Fr

5. Order the following elements by increasing atomic radius:

Be, Mg, Ca, Sr, Ba, Ra

6. Order the following elements by increasing electronegativity:

B, Al, Ga, In, Tl

7. Order the following elements by increasing ionization energy:

Be, Mg, Ca, Sr, Ba, Ra

8. Order the following elements by increasing atomic radius:

Na, K, Rb, Cs, Fr

9. Order the following elements by increasing electronegativity:

F, Cl, Br, I, At

10. Order the following elements by increasing ionization energy:

Li, Na, K, Rb, Cs, Fr

Name: \_\_\_\_\_ Date: \_\_\_\_\_ Period: \_\_\_\_\_

### Periodic Trends Test Study Guide

**Section 1: Atomic Structure and Properties of the Periodic Table**

1. What are the three main categories of elements in the periodic table?

- Metals
- Non-metals
- Metalloids

2. List the following elements in order of increasing atomic radius: Li, Na, K, Rb, Cs, Fr.

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# Periodic Table And Chemistry Study Guide Answers

**Hussin A.Rothana**



## **Periodic Table And Chemistry Study Guide Answers:**

## **Periodic Table And Chemistry Study Guide Answers** Book Review: Unveiling the Power of Words

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liqueur that in most cases is not made with almonds but apricot kernels and peach stones for instance disaronno the most popular and one of the best amaretto brands is made from apricot kernels and burnt sugar

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