

## **Finding the Percent Composition**

- **Percent Composition** - The percentage (by mass) of each element in a compound.

### **Example 1:**

Find the percent composition of red candies in a bag of 200 candies, if there are 26 red, and 174 blue candies.

- Total Candies: 200
- Red Candies: 26
- Blue Candies: 174

$$\frac{26 \text{ red}}{200 \text{ total}} \times 100 = 13.0 \% \text{ red}$$

**Formula:**

$$\frac{\text{Part}}{\text{Total}} \times 100$$

$$\frac{174 \text{ blue}}{200 \text{ total}} \times 100 = 87.0 \% \text{ blue}$$

### **Example 2:**

Find the percent composition of  $\text{MgCl}_2$ .

- Total mass  $\text{MgCl}_2$ : 95.21 g
- Mass of Mg: 24.31 g
- Mass of  $\text{Cl}_2$ : 70.91 g

$$\frac{24.31 \text{ g}}{95.21 \text{ g}} \times 100 = 25.53 \% \text{ Mg}$$

$$\frac{70.91 \text{ g}}{95.21 \text{ g}} \times 100 = 74.47 \% \text{ Cl}_2$$

### **Example 3:**

Find the percent composition of  $\text{H}_2\text{SO}_4$ .

- Total mass  $\text{H}_2\text{SO}_4$ : 98.08 g
- Mass of  $\text{H}_2$ : 2.02 g
- Mass of S: 32.07 g
- Mass of  $\text{O}_4$ : 64.00 g

$$\frac{2.02 \text{ g}}{98.08 \text{ g}} \times 100 = 2.06 \% \text{ H}_2$$

$$\frac{32.07 \text{ g}}{98.08 \text{ g}} \times 100 = 32.70 \% \text{ S}$$

$$\frac{64.00 \text{ g}}{98.08 \text{ g}} \times 100 = 65.25 \% \text{ O}_4$$

# Percent Composition Answers

**S Nieto**



## **Percent Composition Answers:**

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*Solved Problems in Chemistry* Beatriz Elena Soledad Rodríguez, 2010-10-26 This book is written to help you study general chemistry In each subject area will give a brief theoretical explanation of the concepts to study and then find a series of exercises done As you progress through the chapters will encounter increasingly complex problems that will be useful to understand the issues Will be presented a number of problems solved once you understand the explanation try to solve it without seeing the answer to see if fully understood In addition to the problems solved will be given a series of exercises for you to practice and find the answer to every problem It is convenient to solve the exercises in the order given to help in reasoning



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