

ATOMIC STRUCTURE & PERIODIC TABLE QUIZ Name: _____

 K T C
14 6 5

Assessment of Learning (Product)

MULTIPLE CHOICE: Choose the most correct answer. (10 K)



1. The diagram above represents an atom, from its outer boundary to its core. The letter "Y" in the diagram above marks the:

- A) Region of greatest density
- B) location of the protons
- C) location of the neutrons
- D) Nucleus
- E) All of the above

2. In order for an atom to be nitrogen, it must have:

- A) 7 neutrons
- B) a mass of 14.01
- C) 8 protons
- D) an atomic number of 7

3. Which of the following is a TRUE statement about atoms?

- A) All atoms of the same element have the same number of electrons, protons, and neutrons.
- B) Atoms of different elements may have the same number of protons in the nucleus.
- C) All atoms of the same element have the same number of electrons and protons but may have different numbers of neutrons.
- D) The number of neutrons is equal to the number of protons.

4. Different isotopes of a particular element contain different numbers of _____.

- A) protons
- B) neutrons
- C) protons and neutrons
- D) protons, neutrons, and electrons
- E) None of the above is correct.

5. Lithium is a _____ and magnesium is a _____.

- A) nonmetal, metal
- B) nonmetal, nonmetal
- C) metal, metal
- D) metal, metalloid
- E) metalloid, metalloid

6. Iridium, atomic #77, belongs to the

- A) alkaline earth metals
- B) noble gases
- C) metalloids
- D) non-metals
- E) halogens
- F) transition metals
- G) alkali metals

7. Of the elements below, _____ is the most metallic.

- A) sodium
- B) barium
- C) magnesium
- D) calcium
- E) cesium

8. Of the atoms below, _____ is the least electronegative?

- A) Rb
- B) F
- C) Si
- D) Cl
- E) Ca

Pearson Education Atomic Structure Quiz Answers

Michael Seilmaier



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