

Name :

Date :

Introduction to Acid, Base, and Salt

1) Which one of the following is acidic?

- a) Lemon juice b) Tomato c) Milk d) All

Answer ____

2) Which one of the following with turn red litmus blue?

- a) Vinegar b) Baking soda c) Lemon juice d) Soft drinks

Answer ____

3) Which one of the following with turn blue litmus red?

- a) Vinegar b) Lime water
c) Baking soda solution d) Washing soda solution

Answer ____

4) Lime water is

- a) CaO b) Ca(OH)_2 c) CaCO_3 d) CaCl_2

Answer ____

5) A compound that turns lime water into colorless again when excess carbon dioxide is made to pass through it is

- a) Calcium carbonate b) Calcium bicarbonate
c) Calcium oxide d) Calcium hydroxide

Answer ____

6) Which among the following is an alkali oxide?

- a) Carbon dioxide b) Sulfur dioxide c) Sodium oxide d) Nitrogen oxide

Answer ____

7) What gas is liberated when dilute sulfuric acid reacts with zinc granules?

- a) Sulfur dioxide b) Carbon dioxide c) Nitrogen d) Hydrogen

Answer ____

8) A solution reacts with crushed eggshells and releases a gas that turns lime water milky. The solution contains

- a) NaCl b) HCl c) LiCl d) KCl

Answer ____

Pearson Education Acids Bases And Salts Answers

G Thomas



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Prentice Hall Chemistry Antony C. Wilbraham, 2008 Leg Ol Sci Chem Pearson Education, Limited, 2007-10-31

Practical Chemistry Mesay Solomon Tesema, Digafie Zeleke, 2024-11-18 This laboratory manual offers a broad introduction to the chemistry of transition elements and more specifically to the chemistry of titanium vanadium chromium molybdenum manganese iron cobalt nickel copper zinc cadmium and mercury The book includes preparation and properties of these transition metals and introduces the chemistry student to the laboratory skills required for accurate and precise chemical analysis **Longman Active Science 7** Narayanan Vidhu, 2009-09 Lqsg Chemistry O Level 2e ,

Education Outlook , 1898 **Sif: Chemistry 5na Tb** J. G. R. Briggs, 2009 *Principles of Inorganic Chemistry* Brian W. Pfennig, 2015-03-24 Aimed at senior undergraduates and first year graduate students this book offers a principles based approach to inorganic chemistry that unlike other texts uses chemical applications of group theory and molecular orbital theory throughout as an underlying framework This highly physical approach allows students to derive the greatest benefit of topics such as molecular orbital acid base theory band theory of solids and inorganic photochemistry to name a few Takes a principles based group and molecular orbital theory approach to inorganic chemistry The first inorganic chemistry textbook to provide a thorough treatment of group theory a topic usually relegated to only one or two chapters of texts giving it only a cursory overview Covers atomic and molecular term symbols symmetry coordinates in vibrational spectroscopy using the projection operator method polyatomic MO theory band theory and Tanabe Sugano diagrams Includes a heavy dose of group theory in the primary inorganic textbook most of the pedagogical benefits of integration and reinforcement of this material in the treatment of other topics such as frontier MO acid base theory band theory of solids inorganic photochemistry the Jahn Teller effect and Wade s rules are fully realized Very physical in nature compare to other textbooks in the field taking the time to go through mathematical derivations and to compare and contrast different theories of bonding in order to allow for a more rigorous treatment of their application to molecular structure bonding and spectroscopy Informal and engaging writing style worked examples throughout the text unanswered problems in every chapter contains a generous use of informative colorful illustrations *Chemistry: The Central Science* Theodore L. Brown, H. Eugene LeMay Jr., Bruce E. Bursten, Catherine Murphy, Patrick Woodward, Steven Langford, Daluis Sagatys, Adrian George, 2013-10-04 If you think you know the Brown LeMay Bursten Chemistry text think again In response to market request we have created the third Australian edition of the US bestseller Chemistry The Central Science An extensive revision has taken this text to new heights Triple checked for scientific accuracy and consistency this edition is a more seamless and cohesive product yet retains the clarity innovative pedagogy functional problem solving and visuals of the previous version All artwork and images are now consistent in quality across the entire text And with a more traditional and logical organisation of the Organic Chemistry content this comprehensive text is the source of all the information and practice problems students are likely to need for conceptual

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