Heinemann Chemistry 2 Student Workbook 3e

## Chemistry 2 Student Workbook Answers to Worksheets

**Teacher Notes** 

#### UNIT 3

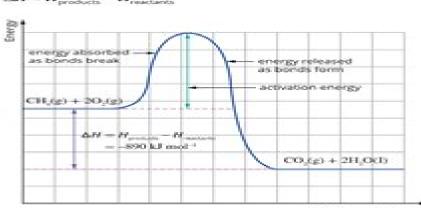
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Worksheet 01 Crossword—exploring energy

ACROSS		DO	DOWN	
3	Coal seam	1	Galvanic	
5	Enthalpy	2	Biofuel	
7	Exothermic	4	Spontaneous	
11	Cathode	6	Biogas	
12	Renewable	8	Joule	
14	Thermochemical	9	Oxidation	
18	Salt bridge	10	Greenhouse	
19	Secondary	13	Activation	
21	Hygroscopic	15	Electrolyte	
22	Fuel	16	Viscosity	
23	Anode	17	Reduction	
		20	Fossil	
		2.2	Fuel	

#### Worksheet 02 Combustion of fuels—thermochemical equations and specific heat capacity

The combustion of fuels releases a significant quantity of energy that can be used as an energy source. Chemical energy is the energy stored in the bonds between atoms and between molecules arising from attractions and repulsions between subatomic particles. It is called enthalpy and is represented by the symbol, H. All combustion reactions are exothermic reactions. In such reactions, the enthalpy change, symbol ΔH, has a negative sign.  $\Delta H = H_{contacts} - H_{contacts}$ 



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**Chao Zhang** 

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