

Covalent Bonding

BIG Idea Covalent bonds form when atoms share electrons.

B.1 The Covalent Bond

MAIN Idea Atoms gain stability when they share electrons and form covalent bonds.

B.2 Naming Molecules

MAIN Idea Specific rules are used when naming binary molecular compounds, binary acids, and oxyacids.

B.3 Molecular Structures

MAIN Idea Structural formulas show the relative positions of atoms within a molecule.

B.4 Molecular Shapes

MAIN Idea The VSEPR model is used to determine molecular shape.

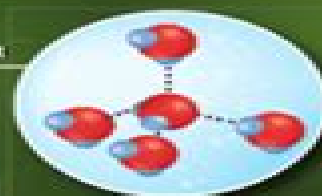
B.5 Electronegativity and Polarity

MAIN Idea A chemical bond's character is related to each atom's attraction for the electrons in the bond.

ChemFacts

- The spherical shape of a water drop is due to surface tension, a phenomenon caused by forces between molecules.
- Surface tension makes water act somewhat like an elastic film. Insects called water striders are able to walk on the filmlike surface of water.
- The chemical and physical properties of water make it a unique liquid.

Spherical water droplet



Space-filling model



Ball-and-stick model



Lewis structure

Pearson Chemistry Foundations Edition Chapter 8 1

Covalent Bonding

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Spratley**



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 - The Rise of Digital Reading Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding
 - Advantages of eBooks Over Traditional Books
2. Identifying Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding
 - Personalized Recommendations
 - Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding User Reviews and Ratings
 - Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding and Bestseller Lists
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 - Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding Budget-Friendly Options
6. Navigating Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding eBook Formats
- ePub, PDF, MOBI, and More
 - Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding Compatibility with Devices
 - Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding Enhanced eBook Features
7. Enhancing Your Reading Experience
- Adjustable Fonts and Text Sizes of Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding
 - Highlighting and Note-Taking Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding
 - Interactive Elements Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding
8. Staying Engaged with Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding
- Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding
9. Balancing eBooks and Physical Books Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding
- Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding
10. Overcoming Reading Challenges
- Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding
- Setting Reading Goals Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding
- Fact-Checking eBook Content of Pearson Chemistry Foundations Edition Chapter 8 1 Covalent Bonding
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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