

ilmkidunya

Student Name	Roll Num	Class Name	Paper Code
Subject Name	Time Allowed	Total Marks	Test Detail
Exam Syllabus	Chapter 2	35	C2

- 1- Circle the correct answer. (8x1=8)**
- Pressure was kept in discharge tube: 4 atm (D) 10^{-4} atm (C) 10^{-2} atm (B) 10 atm (A)
 - Number of electrons used in the discharge tube was: 2 (D) 3 (C) 5 (A) 4 (B)
 - Which one of the following has highest energy value? N (D) M (C) L (B) K (A)
 - How many electrons can accommodate in L shell? 32 (D) 18 (C) 8 (B) 2 (A)
 - N-shell can accommodate electrons: 50 (D) 32 (C) 18 (B) 8 (A)
 - Gamma rays emit from: $^{238}_{92}\text{U}$ (D) Co-60 (C) Sr-90 (B) P-32 (A)
 - Isotopes of chlorine are: 1 (D) 4 (C) 3 (B) 2 (A)
 - Which radioisotope is used for the diagnosis of goiter in thyroid gland? Iodine-131 (B) Cobalt-60 (A) Phosphorus-30 (D) Strontium-90 (C)
- 2- Answer any 4 short questions. (4x2=8)**
- Write the experiments which was done by Sir William Crooks.
 - Define Nucleons.
 - Explain how anode rays are formed from the gas taken in the discharge tube?
 - Why Rutherford considered as the father of nuclear science?
 - Write the electronic configuration of an element having 11 electrons.
 - Magnesium has electronic configuration 2, 8, 2. Why magnesium tend to lose electrons?
- 3- Answer any 5 short questions. (5x2=10)**
- How electrons fill the shell?
 - Write down the electronic configuration of fluorine and Neon.
 - Write down the electronic configuration of Al^{3+} . How many electrons are present in its outermost shell?
 - What are the isotopes of uranium?
 - Where nuclear energy used produced from uranium.
 - Write two isotopes of Uranium and Chlorine.
 - Define Carbon dating.
 - How many protons and neutrons are present in chlorine atom $^{35}_{17}\text{Cl}$?
- Attempt any ONE question in detail. (5+4=9)**
- Compare the difference between the Rutherford's and Neil Bohr's atomic theories.
 - Describe the electronic configuration Na^+ , Mg^{2+} and Al^{3+} ions. Do they have the same number of electrons in the outermost shell?
- 5(a) What is an isotope? Describe the isotopes of hydrogen with diagrams.**
- (b) Write notes on isotopes of Carbon and Chlorine.**

Pearson Chemistry Chap 9 Test

Sabine Zange



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1. Understanding the eBook Pearson Chemistry Chap 9 Test
 - The Rise of Digital Reading Pearson Chemistry Chap 9 Test
 - Advantages of eBooks Over Traditional Books
2. Identifying Pearson Chemistry Chap 9 Test
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pearson Chemistry Chap 9 Test
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pearson Chemistry Chap 9 Test
 - Personalized Recommendations
 - Pearson Chemistry Chap 9 Test User Reviews and Ratings
 - Pearson Chemistry Chap 9 Test and Bestseller Lists
5. Accessing Pearson Chemistry Chap 9 Test Free and Paid eBooks
 - Pearson Chemistry Chap 9 Test Public Domain eBooks
 - Pearson Chemistry Chap 9 Test eBook Subscription Services
 - Pearson Chemistry Chap 9 Test Budget-Friendly Options
6. Navigating Pearson Chemistry Chap 9 Test eBook Formats

- ePub, PDF, MOBI, and More
- Pearson Chemistry Chap 9 Test Compatibility with Devices
- Pearson Chemistry Chap 9 Test Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pearson Chemistry Chap 9 Test
 - Highlighting and Note-Taking Pearson Chemistry Chap 9 Test
 - Interactive Elements Pearson Chemistry Chap 9 Test
- 8. Staying Engaged with Pearson Chemistry Chap 9 Test
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pearson Chemistry Chap 9 Test
- 9. Balancing eBooks and Physical Books Pearson Chemistry Chap 9 Test
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pearson Chemistry Chap 9 Test
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Pearson Chemistry Chap 9 Test
 - Setting Reading Goals Pearson Chemistry Chap 9 Test
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Pearson Chemistry Chap 9 Test
 - Fact-Checking eBook Content of Pearson Chemistry Chap 9 Test
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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