Chapter 19

- a) Contains as many hydrogen atoms as possible for the given number of carbons. Only carboncarbon single bonds.
 - b) (i) C₁₁H₂₄ (ii) liquid (iii) C₁₁H₂₄(i) + 17O₂(g) ---> 11CO₂(g) + 12H₂O(i)
- 2. a) a carbon-carbon double bond

(Home schooling parents: It doesn't matter what position you draw the bromine in on the righthand carbon - N, S or E (with hydrogens in the other two positions). All that matters is that one bromine becomes attached to each of the carbons either side of the original double bond.)

d) (i) CH₄(g) + Br₂(g) ---> CH₃Br(g) + HBr(g)

(ii) In this (substitution) reaction, one of the hydrogens has been replaced by a bromine during the reaction. In the previous (addition) reaction, nothing was lost when the two molecules combined together.

- a) Ethene is passed with steam at 300°C and 60 70 atmospheres over a phosphoric acid catalyst.
 - b) e.g. two of: continuous flow process, therefore faster; rapid reaction; purer product.
 - c) uses renewal resources; gentle conditions.
- 4. Dissolve the sugar in water in a conical flask (or similar) and add yeast. You now need some means of keeping the air out white allowing carbon dioxide to escape. You could use a commercial homewinemaking airlock of some sort, but a plug of cotton wool works just as well. Leave the mixture in the warm for a few days until bubbling stops.



Allow the yeast to settle, and pour off the solution so that it can be fractionally distilled, collecting the fraction which boils at 78°C. (See the diagram for the fractional distillation of an alcohol/water mixture in Chapter 11 of the book.)

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web the 50 reaction turbine has been used widely and has special significance the velocity diagram of a 50 reaction is symmetrical and for the maximum utilization factor the exit velocity v2 must be axial figure 2 40 shows a velocity diagram of a 50 reaction turbine and the effect on the utilization factor

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web p1v1 u1 v12 2 gz1 q p2v2 u2 v22 2 gz2 w p1v1 u1 h1 enthalpy of steam at inlet p2v2 u2 h2 enthalpy of steam at exit h1 v12 2 gz1 q h2 v22 2 gz2 w generally changes in potential energy are negligible z1 z2 if no heat is supplied from the surroundings then q 0

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web velocity diagram fig 7 velocity diagram of pressure compounded reaction turbine the velocity diagram given in figure 7 gives a detail about the various components of steam velocity and blade velocity symbols have the same meaning as above see also pressure compounding in turbines references

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web the above figure shows the velocity and pressure curve while working fluid passing through the blades of velocity compounded turbine here the kinetic energy velocity is extracted from working in multiple stages this is known as velocity compounding figure show the arrangement of a velocity compound turbine

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