

6.2 Classifying the Elements > Electron Configurations in Groups

The Representative Elements Groups 1A through 7A

	1A		2A		3A	4A	5A	6A	7A
1	1 H Hydrogen 1.008								
2	3 Li Lithium 6.941	4 Be Beryllium 9.012			5 B Boron 10.81	6 C Carbon 12.01	7 N Nitrogen 14.01	8 O Oxygen 16.00	9 F Fluorine 18.998
3	11 Na Sodium 22.990	12 Mg Magnesium 24.305			13 Al Aluminum 26.982	14 Si Silicon 28.086	15 P Phosphorus 30.974	16 S Sulfur 32.06	17 Cl Chlorine 35.453
4	19 K Potassium 39.098	20 Ca Calcium 40.078			31 Ga Gallium 69.723	32 Ge Germanium 72.64	33 As Arsenic 74.922	34 Se Selenium 78.96	35 Br Bromine 79.904
5	37 Rb Rubidium 85.468	38 Sr Strontium 87.62			49 In Indium 114.82	50 Sn Tin 118.71	51 Sb Antimony 121.75	52 Te Tellurium 127.6	53 I Iodine 126.905
6	55 Cs Cesium 132.91	56 Ba Barium 137.33			81 Tl Thallium 204.38	82 Pb Lead 207.2	83 Bi Bismuth 208.98	84 Po Polonium 209	85 At Astatine 210
7	87 Fr Francium 223	88 Ra Radium 226							

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Roger Bridgman



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