

Some common compounds are listed below:

silver oxide	lead nitrate	mercury oxide	calcium carbonate	copper oxide
iron oxide	carbon dioxide	sodium chloride	iron sulfide	copper sulfate
	copper carbonate	lead carbonate	lead nitride	

- Which of these compounds:
 - contain all three elements, oxygen, carbon and copper?
 - do not contain any oxygen atoms?
- What elements are in lead nitrate?
 - What is the name of the compound that contains nitrogen and lead only?
- Several of these compounds can take part in thermal decomposition reactions. What happens during a thermal decomposition reaction?
- When copper carbonate is heated it forms two compounds in the above list. One is a metal oxide and the other is a gas.
 - Write a word equation for this reaction?
 - What is the test for the gas formed in this reaction?
- Calcium carbonate is commonly found in limestone, marble and chalk. Limestone is often used to make lime, calcium oxide, which has many uses in agriculture and building.
 - What elements make up calcium carbonate?
 - Write a word equation for the formation of lime from limestone.
 - How could you change calcium carbonate into calcium oxide in the laboratory?
- Look at the word equation below:

silver oxide → silver + oxygen

 - What kind of reaction is this?
 - How is this reaction brought about?
 - Write a similar word equation for mercury oxide reacting in this way.

I can...

- describe examples of thermal decomposition reactions
- name compounds containing two elements plus oxygen
- write chemical word equations.

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