Answers

Chapter 1

Exercises

- (a) CUCO -+CUC + CO
 - (b) 2Mg + O → 2MgO
 - (c) H SO + 2NaOH → Na SO + 2H O
 - (d) N2 " + __ → 2NH3 "
 - (e) CH + 2O → CO + 2H O
- 2 (a)(2K + 2H O → 2KOH + H

QN OH + 30 → 200 + 3H O

COM 210 → 210CH

46003-x20/203 +

F(eD + 3C -+ 3CO + 2Fe

3 (a)(2C,H_ + 13O, → 8CO, + 10H,O

4866. + 50. → 4NO + 6H,O

3(a) + 8HNO → 3CU(NO) + 2NO + 4H O

6(MD + 2N H -+2N + 10H 0 + 0

460 + N + 150 - 800 + 14H 0 + 2N

- 4 (a) Sand and water: heterogeneous
- Sibilities: tretterogeneous.
- Sociar and water: homogeneous
- S(d) and iron filings: heterogeneous
- E(s)and and water; homogeneous
 - (f) Steel: homogeneous
 - 5 (a)(2KNO_(b)) → 2KNO_(b) + O_(g)
- O(60 O₂(s) + H.SO₂(aq) -+ CaSO₂(s) + CO₂(q) + H Offi

264b) + 2H,O() -→ 2L/OH(ac) + H,(a)

P(dNO,Julisq) + 2NaO(laq) → PbOU(s) +

2NeNO,(eq)

(e) 2C,H,(g) + 9O,(g) → 6CO,(g) + 6H,O()

In the answers, exercises and text there is inconsistency about whether the water formed in combustion reactions is gas or liquid. At room temperature it will ultimately finish up as a liquid. ast shown here.

- X has diffused more quickly, so it must be a lighter gas, its particles have greater velocity than the particles of Y at the same temperature. (Note though that they will both have the same value for average kinetic energy.)
- 7 From the kinetic molecular theory we would expect a solid to be more dense than its liquid. and therefore that ice would sink in water.
- 8 Bubbles will be present through the volume of the liquid. A brown gas is visible above the brown liquid. As the two states are at the same temperature, the particles have the same average kinetic energy and are moving at the same speed. The inter-particle distances in the gas are significantly larger than those in the liquid.
- At certain conditions of low temperature and low humidity, snow changes directly to water vapour by sublimation, without going through the liquid chase.
- 10 Steam will condense on the skin, releasing energy as it forms liquid at the same temperature (E-D) on Figure 1.4). This is additional to the energy released when both the boiling water and the condensed steam cool on the surface of the skin.
- 12 AW 01 EX 12

2

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George Carey Foster, Alfred William Porter

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Examination Questions and Answers in Basic Anatomy and Physiology Martin Caon, 2020-08-03 This third edition provides 2900 multiple choice questions on human anatomy and physiology and some biophysical science separated into 20 chapters and 68 categories In addition there are 64 essay topics The answer to each guestion is accompanied by an explanation Each chapter has an introduction to set the scene for the questions to come However not all possible information is provided within these Introductions so an Anatomy and Physiology textbook is an indispensable aid to understanding the answers The textbook offers a more holistic approach to the subjects of anatomy and physiology by also including biomechanics biophysics and biochemistry The questions have been used in end of semester examinations for undergraduate anatomy and physiology courses and as such reflect the focus of these particular courses and are pitched at this level to challenge students that are beginning their training in anatomy and physiology. The guestion and answer combinations are intended for use by teachers to select questions for their next examinations and by students when studying for an upcoming test Students enrolled in the courses for which these questions were written include nursing midwifery paramedic physiotherapy occupational therapy nutrition and dietetics health sciences exercise science and students taking an anatomy and physiology course as an elective The Old Riddle and the Newest Answer John Gerard, 1904 Solution Chemistry Viktor Gutmann, 1995-08-31 This book emphasises those features in solution chemistry which are difficult to measure but essential for the understanding of both the qualitative and the quantitative aspects Attention is paid to the mutual influences between solute and solvent even at extremely small concentrations of the former The described extension of the molecular concept leads to a broad view not by a change in paradigm but by finding the rules for the organizations both at the molecular and the supermolecular level of liquid and solid solutions Inorganic Biochemistry J. A. Cowan, 1997-03-21 The text will provide a set of problems covering mechanistic structural and spectroscopic issues in inorganic chemistry Specific areas to be covered include coordination chemistry physiochemical aspects of solution chemistry inorganic chemistry of biological systems both natural biomolecules and bioinorganic models Illustrative worked examples will be included The problems will be categorized by topic chapters for ease of reference and use in courses They will provide a valuable resource for instructors providing a means of testing and developing the many principles covered in texts and advanced courses Often students find it difficult to find practical problems to test the principles they have learned in class This text will provide a series of questions to test understanding and worked examples as a pedagogical aid The

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