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proton

- -the avey
- -positively charged
- -nucleon

neutron

- -theaver
- -meutrali
- -nucleon

electron

- -very small (1/1800 of proton)
- -negative
- -outside of nucleus

properties of nucleus

- -# of protons determines the element
- -held together by nuclear force
- -# of protons is extremely stable unlike # of electrons

atomic number

equal to the # of protons

atomic mass

of protons + # of neutrons averaged across all of the element on earth

isotopes

- when elements have different masses but the same chemical properties
- -ex. silver-107 and silver-109

lewis dot

- a 2-D model that represents covalent bonds as straight lines, ionic bonds as spaces, and unbonded valence electrons as dots
- -can have single, double, or triple bonds

physical properties of compounds and molecules

- -melting point
- -boilling point
- -solubility
- -electrical conductivity

molecular mass

of atoms(atomic mass)-molecular mass

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G Thomas

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