

Element	symbol	Atomic number	Electronic arrangement in different orbit					
			K	L	M	N	O	P
Hydrogen	H	1	1					
Helium	He	2	2					
Lithium	Li	3	2	1				
Beryllium	Be	4	2	2				
Boron	B	5	2	3				
Carbon	C	6	2	4				
Nitrogen	N	7	2	5				
Oxygen	O	8	2	6				
Fluorine	F	9	2	7				
Neon	Ne	10	2	8				
Sodium	Na	11	2	8	1			
Magnesium	Mg	12	2	8	2			
Aluminium	Al	13	2	8	3			
Silicon	Si	14	2	8	4			
Phosphorus	P	15	2	8	5			
Sulphur	S	16	2	8	6			
Chlorine	Cl	17	2	8	7			
Argon	Ar	18	2	8	8			
Potassium	K	19	2	8	8	1		
Calcium	Ca	20	2	8	8	2		

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● $c = v\lambda$

● $\bar{v} = \frac{1}{\lambda} = \frac{v}{c}$

[c = velocity of light in free space,

v = frequency, λ = wavelength]

[\bar{v} = wave number]

Physical Science Notes Chapter 4 Atomic Structure

N Noddings



Physical Science Notes Chapter 4 Atomic Structure:

Physical Science Notes Chapter 4 Atomic Structure Book Review: Unveiling the Power of Words

In a world driven by information and connectivity, the power of words has been evident than ever. They have the ability to inspire, provoke, and ignite change. Such is the essence of the book **Physical Science Notes Chapter 4 Atomic Structure**, a literary masterpiece that delves deep into the significance of words and their impact on our lives. Written by a renowned author, this captivating work takes readers on a transformative journey, unraveling the secrets and potential behind every word. In this review, we shall explore the book's key themes, examine its writing style, and analyze its overall impact on readers.

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