

Grade 8 Science Notes: Chemical Reactions

Chemical Reactions are processes where substances are transformed into new substances with different properties.

Types of Chemical Reactions:

1. Combination Reaction:

- Two or more substances combine to form one product.
- Example: Hydrogen + Oxygen \rightarrow Water.

2. Decomposition Reaction:

- A compound breaks down into simpler substances.
- Example: Water \rightarrow Hydrogen + Oxygen.

3. Displacement Reaction:

- An element replaces another in a compound.
- Example: Zinc + Copper Sulfate \rightarrow Zinc Sulfate + Copper.

4. Combustion Reaction:

- A substance reacts with oxygen, releasing energy.
- Example: Methane + Oxygen \rightarrow Carbon Dioxide + Water + Energy.

Reactants and Products:

- Reactants: Starting substances.
- Products: New substances formed.

Indicators of a Chemical Reaction:

1. Color Change.
2. Gas Production (bubbles).
3. Temperature Change (heat release/absorption).
4. Formation of Precipitate.

Importance:

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Clive Lon



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Decoding **Physical Science Grade8 Chemical Reaction Pearson**: Revealing the Captivating Potential of Verbal Expression

In a period characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its capability to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Physical Science Grade8 Chemical Reaction Pearson**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers attempt an enlightening odyssey, unraveling the intricate significance of language and its enduring affect our lives. In this appraisal, we shall explore the book is central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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