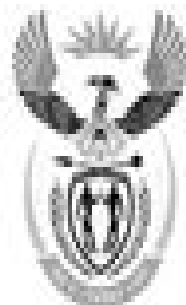


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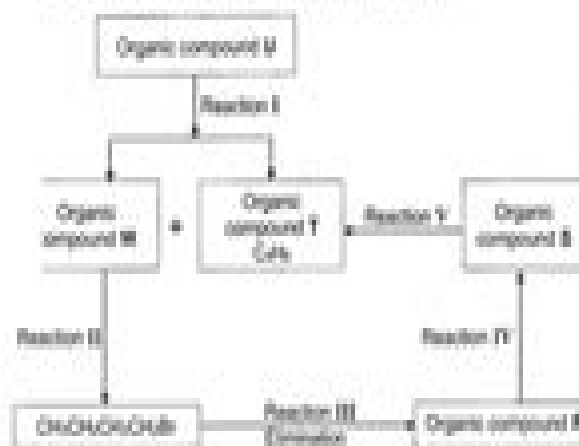
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Diagram on page 9 is redrawn below for easy reference.



Changes to the conditions of equilibrium are made at different times. The graph shows the results obtained. X, Y and Z represent the gases in the above reaction.



2.1.1 Define for both chemical equilibrium

Use the graph to answer the questions below.

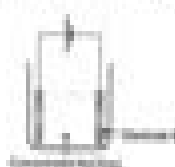
The inorganic reagents shown below are available for reactions IV and Y.

Br ₂	H ₂ SO ₄ (conc.)	NaOH(aq.)	HBr	H ₂
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4.5 Write down the TYPE of elimination in reaction III.

4.6 Compounds R and T are positional isomers.

4.7 Identify the reaction in the above diagram that is a redox reaction.



4.8 Write the half-cell reaction for the anode in the above cell.

4.9 Write the overall cell reaction for the above cell.

5.1 Write down the name of the reaction in the above diagram.

5.2 Write down the name of the reaction in the above diagram.



5.3 Write down the name of the reaction in the above diagram.

5.4 Write down the name of the reaction in the above diagram.

5.5 Write down the name of the reaction in the above diagram.

5.6 Write down the name of the reaction in the above diagram.

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1. Understanding the eBook Physical Science Final Grade Paper November 2014
 - The Rise of Digital Reading Physical Science Final Grade Paper November 2014
 - Advantages of eBooks Over Traditional Books
2. Identifying Physical Science Final Grade Paper November 2014
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Physical Science Final Grade Paper November 2014
 - User-Friendly Interface
4. Exploring eBook Recommendations from Physical Science Final Grade Paper November 2014
 - Personalized Recommendations
 - Physical Science Final Grade Paper November 2014 User Reviews and Ratings
 - Physical Science Final Grade Paper November 2014 and Bestseller Lists
5. Accessing Physical Science Final Grade Paper November 2014 Free and Paid eBooks
 - Physical Science Final Grade Paper November 2014 Public Domain eBooks
 - Physical Science Final Grade Paper November 2014 eBook Subscription Services

- Physical Science Final Grade Paper November 2014 Budget-Friendly Options
- 6. Navigating Physical Science Final Grade Paper November 2014 eBook Formats
 - ePub, PDF, MOBI, and More
 - Physical Science Final Grade Paper November 2014 Compatibility with Devices
 - Physical Science Final Grade Paper November 2014 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Physical Science Final Grade Paper November 2014
 - Highlighting and Note-Taking Physical Science Final Grade Paper November 2014
 - Interactive Elements Physical Science Final Grade Paper November 2014
- 8. Staying Engaged with Physical Science Final Grade Paper November 2014
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Physical Science Final Grade Paper November 2014
- 9. Balancing eBooks and Physical Books Physical Science Final Grade Paper November 2014
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Physical Science Final Grade Paper November 2014
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Physical Science Final Grade Paper November 2014
 - Setting Reading Goals Physical Science Final Grade Paper November 2014
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Physical Science Final Grade Paper November 2014
 - Fact-Checking eBook Content of Physical Science Final Grade Paper November 2014
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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