



NATIONAL  
SENIOR CERTIFICATE

GRADE 11

NOVEMBER 2020

PHYSICAL SCIENCES P2  
(CHEMISTRY)  
EXEMPLAR

MARKS: 100  
TIME: 1 hour



The question paper consists of 10 pages, including 4 extra sheets.  
Use permanent marker for writing and drawing.

## QUESTION 3

### PHYSICAL SCIENCES PAPER 2

### INTERMOLECULAR FORCES

**GRADE 11**

**NOVEMBER 2020**

Examiner's use: \_\_\_\_\_ Physical Sciences P2

QUESTION 3 starts on a NEW page.

- 3.1 The boiling points in the table below were obtained during an investigation carried out to measure the boiling points of pure 1-hexanol. The table also lists the  $\Delta H_{\text{vap}}$  and  $\Delta H_{\text{fusion}}$  for 1-hexanol in kJ mol<sup>-1</sup> and 1 mol of 1-hexanol.

| Compound | Molecular mass (M <sub>r</sub> ) | Boiling point (°C) |
|----------|----------------------------------|--------------------|
| A        | 102                              | 16,2               |
| B        | 102                              | 21                 |

- 3.1.1 Define the term boiling point. (2)
- 3.1.2 Is pure 1-hexanol only composed of A and B, or A, B and C, and D, one of which is present? (2)
- 3.1.3 Name the type of intermolecular force that exists between molecules of both compounds A and B and is the prime reason of their volatility. (2)
- 3.1.4 What (kind of) intermolecular force(s) is/are (strongest) present between 1-hexanol molecules? (2)
- 3.1.5 Which compound (A or B) will have the higher vapour pressure? (2)
- 3.1.6 Give a reason for your answer. (2)
- 3.1.7 Which compound (A or B) will have the higher surface tension? (2)
- 3.1.8 Give a reason for the answer to 3.1.7 relating to the space the table. (2)

Compound C (M<sub>r</sub> 102) will show in the table, has a smaller molecular mass than both compounds A and B but has a relatively higher boiling point of 16,2 °C.

- 3.1.8 Explain why the boiling point of compound C is (MAYBE) higher than that of compound A and B by referring to the  $\Delta H_{\text{vap}}$  and intermolecular or intramolecular forces present. (2)

- 3.2 Substances D and E are added to equal volumes of pure 1-hexanol (100 mL) and water. It is observed that when the substances are added, the liquid volume



- 3.2.1 Is either substance D or E, or both, ionic in character? (2)
- 3.2.2 Explain the answer to QUESTION 3.2.1 about the solution in the intermolecular forces involved. (2)

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# Physical Science Paper2 November 2014

**J Dewey**



**Physical Science Paper2 November 2014:**

## Decoding **Physical Science Paper2 November 2014**: Revealing the Captivating Potential of Verbal Expression

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