

# Memorandum Grade 11 p/sciences

Physical Sciences Gr11/P2

2  
NSC

September

## QUESTION 1

1.1 A ✓✓

(2)

1.2 B ✓✓

(2)

1.3 C & ✓✓

(2)

1.4 B & ✓✓

(2)

1.5 C & ✓✓

(2)

1.6 D ✓✓

(2) ~~(10)~~ (12)

## QUESTION 2

2.1 The mass of one mole of a substance measured in  $\text{g} \cdot \text{mol}^{-1}$  ✓✓ (2 or 0)

(2)

2.2

| Marking criteria                                                                                             |                                                                           |                           |
|--------------------------------------------------------------------------------------------------------------|---------------------------------------------------------------------------|---------------------------|
| (a) Any formula $m = nM$ or $N = nN_A$ ✓                                                                     |                                                                           |                           |
| (b) Substitution of $3,114 \times 10^{24}$ and $6,02 \times 10^{23}$ into $N = nN_A$ ✓                       |                                                                           |                           |
| (c) Substitution of $1 \text{ g} \cdot \text{mol}^{-1}$ into $m = nM$ ✓                                      |                                                                           |                           |
| (d) Mole ratio 3:6:1 ✓                                                                                       |                                                                           |                           |
| (e) $X = 3$ ✓, $Y = 6$ ✓, $Z = 1$ ✓                                                                          |                                                                           |                           |
| C                                                                                                            | H                                                                         | O                         |
| $n = \frac{N}{N_A}$ ✓<br>$n = \frac{3,114 \times 10^{24}}{6,02 \times 10^{23}}$ ✓<br>$n = 5,173 \text{ mol}$ | $n = \frac{m}{M}$<br>$n = \frac{10,345}{1}$ ✓<br>$n = 10,345 \text{ mol}$ | $n = 1,724 \text{ mol}$   |
| $\frac{5,173}{1,724} = 3$                                                                                    | $\frac{10,345}{1,724} = 6$                                                | $\frac{1,724}{1,724} = 1$ |
| 3:6:1 ✓                                                                                                      |                                                                           |                           |
| $X = 3$ ✓                                                                                                    | $Y = 6$ ✓                                                                 | $O = 1$ ✓                 |

Accept values indicated in the empirical formula:  $\text{C}_3\text{H}_6\text{O}$  ✓✓✓

(7)

# Physical Science Memoradam Grade 11 September 2013

**N Noddings**



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