

# Periodic Table of the Elements

1 IA <b>H</b> Hydrogen (1.008)																	18 VIIIA <b>He</b> Helium (4.003)				
3 IIA <b>Li</b> Lithium (6.941)	4 IIA <b>Be</b> Beryllium (9.012)															5 IIIA <b>B</b> Boron (10.81)	6 IIIA <b>C</b> Carbon (12.011)	7 IVA <b>N</b> Nitrogen (14.007)	8 IVA <b>O</b> Oxygen (15.999)	9 VA <b>F</b> Fluorine (18.998)	10 VA <b>Ne</b> Neon (20.180)
11 IA <b>Na</b> Sodium (22.990)	12 IIA <b>Mg</b> Magnesium (24.305)															13 IIIA <b>Al</b> Aluminum (26.982)	14 IIIA <b>Si</b> Silicon (28.086)	15 IVA <b>P</b> Phosphorus (30.974)	16 IVA <b>S</b> Sulfur (32.06)	17 VA <b>Cl</b> Chlorine (35.45)	18 VA <b>Ar</b> Argon (39.948)
19 IA <b>K</b> Potassium (39.098)	20 IIA <b>Ca</b> Calcium (40.078)	21 IIB <b>Sc</b> Scandium (44.956)	22 IIB <b>Ti</b> Titanium (47.88)	23 IIB <b>V</b> Vanadium (50.942)	24 IIB <b>Cr</b> Chromium (51.996)	25 IIB <b>Mn</b> Manganese (54.938)	26 IIB <b>Fe</b> Iron (55.845)	27 IIB <b>Co</b> Cobalt (58.933)	28 IIB <b>Ni</b> Nickel (58.693)	29 IB <b>Cu</b> Copper (63.546)	30 IB <b>Zn</b> Zinc (65.38)	31 IIB <b>Ga</b> Gallium (69.723)	32 IIB <b>Ge</b> Germanium (72.64)	33 IIIA <b>As</b> Arsenic (74.922)	34 IIIA <b>Se</b> Selenium (78.96)	35 IIIA <b>Br</b> Bromine (79.904)	36 IIIA <b>Kr</b> Krypton (83.798)				
37 IA <b>Rb</b> Rubidium (85.468)	38 IIA <b>Sr</b> Strontium (87.62)	39 IIB <b>Y</b> Yttrium (88.906)	40 IIB <b>Zr</b> Zirconium (91.224)	41 IIB <b>Nb</b> Niobium (92.906)	42 IIB <b>Mo</b> Molybdenum (95.94)	43 IIB <b>Tc</b> Technetium (98.906)	44 IIB <b>Ru</b> Ruthenium (101.07)	45 IIB <b>Rh</b> Rhodium (102.91)	46 IIB <b>Pd</b> Palladium (106.36)	47 IB <b>Ag</b> Silver (107.87)	48 IB <b>Cd</b> Cadmium (112.41)	49 IIB <b>In</b> Indium (114.82)	50 IIB <b>Sn</b> Tin (118.71)	51 IIIA <b>Sb</b> Antimony (121.76)	52 IIIA <b>Te</b> Tellurium (127.6)	53 IIIA <b>I</b> Iodine (126.91)	54 IIIA <b>Xe</b> Xenon (131.29)				
55 IA <b>Cs</b> Cesium (132.91)	56 IIA <b>Ba</b> Barium (137.33)	57-71 Lanthanides	72 IIB <b>Hf</b> Hafnium (178.49)	73 IIB <b>Ta</b> Tantalum (180.95)	74 IIB <b>W</b> Tungsten (183.84)	75 IIB <b>Re</b> Rhenium (186.21)	76 IIB <b>Os</b> Osmium (190.23)	77 IIB <b>Ir</b> Iridium (192.22)	78 IIB <b>Pt</b> Platinum (195.08)	79 IB <b>Au</b> Gold (196.97)	80 IB <b>Hg</b> Mercury (200.59)	81 IIB <b>Tl</b> Thallium (204.38)	82 IIB <b>Pb</b> Lead (207.2)	83 IIIA <b>Bi</b> Bismuth (208.98)	84 IIIA <b>Po</b> Polonium (209)	85 IIIA <b>At</b> Astatine (210)	86 IIIA <b>Rn</b> Radon (222)				
87 IA <b>Fr</b> Francium (223)	88 IIA <b>Ra</b> Radium (226)	89-103 Actinides	104 IIB <b>Rf</b> Rutherfordium (261)	105 IIB <b>Db</b> Dubnium (262)	106 IIB <b>Sg</b> Seaborgium (266)	107 IIB <b>Bh</b> Bohrium (264)	108 IIB <b>Hs</b> Hassium (277)	109 IIB <b>Mt</b> Meitnerium (268)	110 IIB <b>Ds</b> Darmstadtium (271)	111 IB <b>Rg</b> Roentgenium (272)	112 IB <b>Cn</b> Copernicium (285)	113 IIB <b>Nh</b> Nihonium (284)	114 IIB <b>Fl</b> Flerovium (289)	115 IIIA <b>Mc</b> Moscovium (288)	116 IIIA <b>Lv</b> Livermorium (293)	117 IIIA <b>Ts</b> Tennessine (289)	118 IIIA <b>Og</b> Oganesson (294)				

57 La Lanthanum (138.91)	58 Ce Cerium (140.12)	59 Pr Praseodymium (140.91)	60 Nd Neodymium (144.24)	61 Pm Promethium (145)	62 Sm Samarium (150.36)	63 Eu Europium (151.96)	64 Gd Gadolinium (157.25)	65 Tb Terbium (158.93)	66 Dy Dysprosium (162.50)	67 Ho Holmium (164.93)	68 Er Erbium (167.26)	69 Tm Thulium (168.93)	70 Yb Ytterbium (173.05)	71 Lu Lutetium (174.967)
89 Ac Actinium (227)	90 Th Thorium (232.04)	91 Pa Protactinium (231.04)	92 U Uranium (238.03)	93 Np Neptunium (237.05)	94 Pu Plutonium (244.06)	95 Am Americium (243.06)	96 Cm Curium (247.07)	97 Bk Berkelium (247.07)	98 Cf Californium (251.08)	99 Es Einsteinium (252.08)	100 Fm Fermium (257.10)	101 Md Mendelevium (258.11)	102 No Nobelium (259.10)	103 Lr Lawrencium (262.11)

# Periodic Table Section 3

**Padhraic Smyth**

A red circular graphic with a gradient, appearing as a stylized arrow or a partial circle, located to the right of the author's name.

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**Structuralist Knowledge Representation** , 2022-05-16

**Organic Chemistry** Michael B. Smith, 2011-06-29 Based on the premise that many if not most reactions in organic chemistry can be explained by variations of fundamental acid base concepts Organic Chemistry An Acid Base Approach provides a framework for understanding the subject that goes beyond mere memorization The individual steps in many important mechanisms rely on acid base reactions and the ability to see these relationships makes understanding organic chemistry easier Using several techniques to develop a relational understanding this textbook helps

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**Physical Science in the Modern World** Jerry Marion, 2012-12-02 Physical Science in the Modern World surveys the whole range of the non biological sciences This book explores the significant ideas and concepts in chemistry physics astronomy geology and meteorology with emphasis on how these sciences bear strongly upon one another and how the basic principles are applied to each Organized into three part encompassing 29 chapters this book starts with an overview of the fundamental building blocks of matter and explains how they are assembled to form molecules rocks minerals and the Earth This text then examines the basic concepts of physical science by exploring the fundamental principles that govern all physical processes and we see how they relate to various everyday occurrences Other chapters consider how modern chemistry affects the world we live in and explain how the development of semiconductor materials has led in the development of miniature electronics This book is a valuable resource for physicists chemists astronomers geologists and meteorologists

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