



Source: The Public Health Service's 1997 Morbidity Tables and related data.

67	La	Lanthanum	138.91
68	Ce	Cerium	140.12
69	Pr	Praseodymium	140.91
70	Nd	Niobium	144.24
71	Pm	Promethium	144.91
72	Sm	Samarium	150.36
73	Eu	Europium	151.96
74	Gd	Gadolinium	157.25
75	Tb	Terbium	158.93
76	Dy	Dysprosium	162.50
77	Ho	Holmium	164.93
78	Er	Erbium	167.26
79	Tm	Thulium	168.93
80	Yb	Ytterbium	173.05
81	Lu	Lutetium	174.97
82	Ac	Actinium	227.03
83	Th	Thorium	232.04
84	Pa	Protactinium	231.04
85	U	Uranium	238.03
86	Np	Neptunium	237.05
87	Pu	Plutonium	244.06
88	Am	Americium	243.06
89	Cm	Curium	247.07
90	Bk	Berkelium	247.07
91	Cf	Californium	251.08
92	Es	Einsteinium	252.08
93	Fm	Fermium	257.10
94	Md	Mendelevium	258.10
95	No	Nobelium	259.10
96	Lr	Lanthanum	262.11

Hydrogen is a chemical element with chemical symbol H and atomic number 1. With an atomic weight of 1.00794 u, hydrogen is the lightest element on the periodic table. Its monatomic form (H) is the most abundant chemical substance in the universe, constituting roughly 75% of all baryonic mass. Non-remnant stars are mainly composed of hydrogen in its plasma state. The most common isotope of hydrogen, termed protium (name rarely used, symbol ¹H), has a single proton and zero neutrons.

Helium is a chemical element with symbol He and atomic number 2. It is a colorless, odorless, tasteless, non-toxic, inert, monatomic gas that heads the noble gas group in the periodic table. Its boiling and melting points are the lowest among the elements and it exists only as a gas except in extreme conditions.

Lithium (from Greek: λίθος lithos, "stone") is a chemical element with symbol Li and atomic number 3. It is a soft, silver-white metal belonging to the alkali metal group of chemical elements. Under standard conditions it is the lightest metal and the least dense solid element. Like all alkali metals, lithium is highly reactive and flammable. For this reason, it is typically stored in mineral oil. When cut open, lithium exhibits a metallic luster, but contact with moist air corrodes the surface quickly to a dull silvery gray, then black tarnish. Because of its high reactivity, lithium never occurs freely in nature, and instead, only appears in compounds, which are usually ionic. Lithium occurs in a number of pegmatitic minerals, but due to its solubility as an ion is present in ocean water and is commonly obtained from brines and clays.

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