

Directions: Choose the letter that corresponds to the correct answer.

1. Which of the following can be considered as Dobereiner's triad?
 - a. He: 4.0, Li: 6.9, Na: 23.0
 - b. N: 14.0, P: 40.0, At: 217.0
 - c. Ne: 20.2, Ar: 39.9, Kr: 83.8
 - d. Ca: 40.1, Sr: 87.6, Ba: 137.3
2. Which of the following elements does not belong to the group?
 - a. N
 - b. S
 - c. Po
 - d. Te
3. Which of the following is/are TRUE about boron and oxygen?
 - a. both are metalloids
 - b. boron is smaller than oxygen
 - c. boron is more electronegative than oxygen
 - d. higher energy is required to eject an electron from the valence shell of boron
 - e. all of the above
 - f. none of the above

For 4 and 5, consider Na, Mg, Al, and Si.

4. The element that ejects one electron from its valence shell most readily is _____.
 - a. Na
 - b. Mg
 - c. Al
 - d. Si
5. Which element has the most negative electron affinity?
 - e. Na
 - f. Mg

Periodic Trends Practice Questions

David Baud



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Table of Contents Periodic Trends Practice Questions

1. Understanding the eBook Periodic Trends Practice Questions
 - The Rise of Digital Reading Periodic Trends Practice Questions
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Trends Practice Questions
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Trends Practice Questions
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Trends Practice Questions
 - Personalized Recommendations
 - Periodic Trends Practice Questions User Reviews and Ratings
 - Periodic Trends Practice Questions and Bestseller Lists
5. Accessing Periodic Trends Practice Questions Free and Paid eBooks
 - Periodic Trends Practice Questions Public Domain eBooks
 - Periodic Trends Practice Questions eBook Subscription Services
 - Periodic Trends Practice Questions Budget-Friendly Options

6. Navigating Periodic Trends Practice Questions eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Trends Practice Questions Compatibility with Devices
 - Periodic Trends Practice Questions Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Trends Practice Questions
 - Highlighting and Note-Taking Periodic Trends Practice Questions
 - Interactive Elements Periodic Trends Practice Questions
8. Staying Engaged with Periodic Trends Practice Questions
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Periodic Trends Practice Questions
9. Balancing eBooks and Physical Books Periodic Trends Practice Questions
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Trends Practice Questions
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Periodic Trends Practice Questions
 - Setting Reading Goals Periodic Trends Practice Questions
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Periodic Trends Practice Questions
 - Fact-Checking eBook Content of Periodic Trends Practice Questions
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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