

PERIODIC TABLE (ADVANCED)

Molecular Structure of Hydrogen



Basic Information

Name: Hydrogen

Symbol: H

Atomic Number: 1

Atomic Mass: 1.00794 amu

Melting Point: -259.14°C ($14.009185^{\circ}\text{K}$, $-434.45203^{\circ}\text{F}$)

Boiling Point: -252.87°C ($20.280105^{\circ}\text{K}$, -423.166°F)

Number of Protons/Electrons: 1

Number of Neutrons: 0

Classification: Non-metal

Crystal Structure: Hexagonal

Density @ 293°K : 0.000089 g/cm^3

Color: colorless

Atomic Structure

Number of Energy Levels: 1

First Energy Level: 1

Isotopes

Isotope	Half-life
H-1	Stable
H-2	Stable
H-3	12.3 years

Facts

Name Origin: Greek word *hydro* (water) and *genes* (generated)

Uses: Balloons, metal refining

Obtained From: mines, oil, gas wells

Molecular Structure of Carbon



Name: Carbon

Symbol: C

Atomic Number: 6

Atomic Mass: 12.0107 amu

Melting Point: 3500.0°C (3773.15°K , 6332.0°F)

Boiling Point: 4827.0°C (5100.15°K , 8700.6°F)

Number of Protons/Electrons: 6

Number of Neutrons: 6

Classification: Non-metal

Crystal Structure: Hexagonal

Density @ 293°K : 2.62 g/cm^3

Color: May be black

Atomic Structure

Number of Energy Levels: 2

First Energy Level: 2

Second Energy Level: 4

Isotopes

Isotope	Half-life
C-11	20.3 minutes
C-12	Stable
C-13	Stable
C-14	5730.0 years
C-15	2.3 seconds

Facts

Uses: steel, fibers

Obtained From: burning with insufficient oxygen

Periodic Table Advanced Speedy Study Guide

Joseph L. Jorden, Dandy Weyn



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table of elements displays not only the elements but the ions that form each element A pamphlet with such a visual aid would greatly benefit chemistry students Any student taking chemistry will need to learn the elements A pamphlet would be concise and break the information down simply making it easier to understand and remember It allows students to simply focus on the main point rather than taking in information that they may or may not need **ASAP Chemistry: a Quick-Review**

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Table of Contents Periodic Table Advanced Speedy Study Guide

1. Understanding the eBook Periodic Table Advanced Speedy Study Guide
 - The Rise of Digital Reading Periodic Table Advanced Speedy Study Guide
 - Advantages of eBooks Over Traditional Books
2. Identifying Periodic Table Advanced Speedy Study Guide
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Periodic Table Advanced Speedy Study Guide
 - User-Friendly Interface
4. Exploring eBook Recommendations from Periodic Table Advanced Speedy Study Guide
 - Personalized Recommendations
 - Periodic Table Advanced Speedy Study Guide User Reviews and Ratings
 - Periodic Table Advanced Speedy Study Guide and Bestseller Lists
5. Accessing Periodic Table Advanced Speedy Study Guide Free and Paid eBooks
 - Periodic Table Advanced Speedy Study Guide Public Domain eBooks
 - Periodic Table Advanced Speedy Study Guide eBook Subscription Services
 - Periodic Table Advanced Speedy Study Guide Budget-Friendly Options
6. Navigating Periodic Table Advanced Speedy Study Guide eBook Formats
 - ePub, PDF, MOBI, and More
 - Periodic Table Advanced Speedy Study Guide Compatibility with Devices
 - Periodic Table Advanced Speedy Study Guide Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Periodic Table Advanced Speedy Study Guide
 - Highlighting and Note-Taking Periodic Table Advanced Speedy Study Guide
 - Interactive Elements Periodic Table Advanced Speedy Study Guide
8. Staying Engaged with Periodic Table Advanced Speedy Study Guide

- Joining Online Reading Communities
- Participating in Virtual Book Clubs
- Following Authors and Publishers Periodic Table Advanced Speedy Study Guide
- 9. Balancing eBooks and Physical Books Periodic Table Advanced Speedy Study Guide
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Periodic Table Advanced Speedy Study Guide
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Periodic Table Advanced Speedy Study Guide
 - Setting Reading Goals Periodic Table Advanced Speedy Study Guide
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Periodic Table Advanced Speedy Study Guide
 - Fact-Checking eBook Content of Periodic Table Advanced Speedy Study Guide
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
- 14. Embracing eBook Trends
 - Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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