



## Percent Yield

45 grams of nitrogen monoxide (NO) react with an excess of sodium hydroxide (NaOH) following the balanced equation above. 6 grams of water were obtained from the reaction

What is the % yield?

Nitrogen monoxide (NO) gmm:

$$(1)(14.0) + (1)(16.0) = \textcircled{30.0 \text{ g/mol}}$$

$$\text{Actual Yield} = 6 \text{ g H}_2\text{O}$$

Water (H<sub>2</sub>O) gmm:

$$(2)(1.0) + (1)(16.0) = \textcircled{18.0 \text{ g/mol}}$$

$$\text{Theoretical Yield} = 6.75 \text{ g H}_2\text{O}$$

$$\frac{\text{Actual Yield}}{\text{Theoretical Yield}} \times 100 = \textcircled{88.9\%}$$

$$\frac{45 \cancel{\text{g NO}} \mid 1 \cancel{\text{mol NO}} \mid 1 \cancel{\text{mol H}_2\text{O}} \mid 18 \text{ g H}_2\text{O}}{30 \cancel{\text{g NO}} \mid 4 \cancel{\text{mol NO}} \mid 1 \cancel{\text{mol H}_2\text{O}}} = 6.75 \text{ g H}_2\text{O}$$

# Percent Yield Problems Answers

**Wolfgang Guggemos**

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

## **Percent Yield Problems Answers:**

## Decoding **Percent Yield Problems Answers**: Revealing the Captivating Potential of Verbal Expression

In an era characterized by interconnectedness and an insatiable thirst for knowledge, the captivating potential of verbal expression has emerged as a formidable force. Its power to evoke sentiments, stimulate introspection, and incite profound transformations is genuinely awe-inspiring. Within the pages of "**Percent Yield Problems Answers**," a mesmerizing literary creation penned by way of a celebrated wordsmith, readers set about an enlightening odyssey, unraveling the intricate significance of language and its enduring impact on our lives. In this appraisal, we shall explore the book's central themes, evaluate its distinctive writing style, and gauge its pervasive influence on the hearts and minds of its readership.

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