

Chapter 1: Limiting Reagents and Percent Yield (Pages 1-3)

Remember: Limiting reagent is the reactant that will completely run out first in a chemical reaction.

Excess: All reagents except the limiting reagent will be present in excess. Remember, limiting reagent is the reactant that will completely run out first in a chemical reaction.

Remember: Limiting reagent is the reactant that will completely run out first in a chemical reaction. Excess: All reagents except the limiting reagent will be present in excess.



The ratio of the limiting reagent to the excess reagent is 2:1.

| | |
|------------------|--------|
| 2H ₂ | 20.0 g |
| O ₂ | 10.0 g |
| H ₂ O | 20.0 g |
| H ₂ | 10.0 g |

1. 20.0 g of H₂ and 10.0 g of O₂ are mixed. 20.0 g of H₂O is produced. How much O₂ is left over?



2. 20.0 g of H₂ and 10.0 g of O₂ are mixed. 20.0 g of H₂O is produced. How much O₂ is left over?

3. 20.0 g of H₂ and 10.0 g of O₂ are mixed. 20.0 g of H₂O is produced. How much O₂ is left over?



4. 20.0 g of H₂ and 10.0 g of O₂ are mixed. 20.0 g of H₂O is produced. How much O₂ is left over?



Percent Yield And Limiting Reactants Quiz

Kaplan Test Prep

A decorative graphic consisting of a light blue horizontal bar with a rounded right end, and a red circular shape partially visible behind it.

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