

5. A compound has the following percentage composition: 74.1% Oxygen and 5.9% Hydrogen. The molecular weight of this compound is 34. What is the compound's true formula?

6. A compound has the following percentage composition: 54.5% Carbon, 13.6% Hydrogen, and 31.9% Nitrogen. The molecular weight of this compound is 60. What is the compound's true formula?

# Percent Composition And Chemical Formulas Quiz

**G. William Daub, William S. Seese**



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