

Answers

Chapter 1

Exercises

- 1 (a) $\text{CuCO}_3 \rightarrow \text{CuO} + \text{CO}_2$
 (b) $2\text{Mg} + \text{O}_2 \rightarrow 2\text{MgO}$
 (c) $\text{H}_2\text{SO}_4 + 2\text{NaOH} \rightarrow \text{Na}_2\text{SO}_4 + 2\text{H}_2\text{O}$
 (d) $\text{N}_2 + 3\text{H}_2 \rightarrow 2\text{NH}_3$
 (e) $\text{CH}_4 + 2\text{O}_2 \rightarrow \text{CO}_2 + 2\text{H}_2\text{O}$
- 2 (a) $2\text{K} + 2\text{H}_2\text{O} \rightarrow 2\text{KOH} + \text{H}_2$
 (b) $\text{C}_2\text{H}_5\text{OH} + 3\text{O}_2 \rightarrow 2\text{CO}_2 + 3\text{H}_2\text{O}$
 (c) $\text{Cl}_2 + 2\text{KI} \rightarrow 2\text{KCl} + \text{I}_2$
 (d) $4\text{CrO}_2 \rightarrow 2\text{Cr}_2\text{O}_3 + 3\text{O}_2$
 (e) $\text{Fe}_2\text{O}_3 + 3\text{C} \rightarrow 3\text{CO} + 2\text{Fe}$
- 3 (a) $2\text{C}_2\text{H}_6 + 13\text{O}_2 \rightarrow 8\text{CO}_2 + 10\text{H}_2\text{O}$
 (b) $4\text{NH}_3 + 5\text{O}_2 \rightarrow 4\text{NO} + 6\text{H}_2\text{O}$
 (c) $3\text{Cu} + 8\text{HNO}_3 \rightarrow 3\text{Cu}(\text{NO}_3)_2 + 2\text{NO} + 4\text{H}_2\text{O}$
 (d) $6\text{H}_2\text{O}_2 + 2\text{N}_2\text{H}_4 \rightarrow 2\text{N}_2 + 10\text{H}_2\text{O} + \text{O}_2$
 (e) $4\text{C}_3\text{H}_7\text{N} + 15\text{O}_2 \rightarrow 8\text{CO}_2 + 14\text{H}_2\text{O} + 2\text{N}_2$
- 4 (a) Sand and water: heterogeneous
 (b) Smoke: heterogeneous
 (c) Sugar and water: homogeneous
 (d) Salt and iron filings: heterogeneous
 (e) Ethanol and water: homogeneous
 (f) Steel: homogeneous
- 5 (a) $2\text{KNO}_3(\text{s}) \rightarrow 2\text{KNO}_2(\text{s}) + \text{O}_2(\text{g})$
 (b) $\text{CaCO}_3(\text{s}) + \text{H}_2\text{SO}_4(\text{aq}) \rightarrow \text{CaSO}_4(\text{s}) + \text{CO}_2(\text{g}) + \text{H}_2\text{O}(\text{l})$
 (c) $2\text{Li}(\text{s}) + 2\text{H}_2\text{O}(\text{l}) \rightarrow 2\text{LiOH}(\text{aq}) + \text{H}_2(\text{g})$
 (d) $\text{Pb}(\text{NO}_3)_2(\text{aq}) + 2\text{NaCl}(\text{aq}) \rightarrow \text{PbCl}_2(\text{s}) + 2\text{NaNO}_3(\text{aq})$
 (e) $2\text{C}_2\text{H}_6(\text{g}) + 9\text{O}_2(\text{g}) \rightarrow 6\text{CO}_2(\text{g}) + 6\text{H}_2\text{O}(\text{l})$

In the answers, exercises and text there is inconsistency about whether the water formed in combustion reactions is gas or liquid. At room temperature it will ultimately finish up as a liquid, as shown here.

- 6 X has diffused more quickly, so it must be a lighter gas. Its particles have greater velocity than the particles of Y at the same temperature. (Note though that they will both have the same value for average kinetic energy.)
- 7 From the kinetic molecular theory we would expect a solid to be more dense than its liquid, and therefore that ice would sink in water.
- 8 Bubbles will be present through the volume of the liquid. A brown gas is visible above the brown liquid. As the two states are at the same temperature, the particles have the same average kinetic energy and are moving at the same speed. The inter-particle distances in the gas are significantly larger than those in the liquid.
- 9 At certain conditions of low temperature and low humidity, snow changes directly to water vapour by sublimation, without going through the liquid phase.
- 10 Steam will condense on the skin, releasing energy as it forms liquid at the same temperature (E–D on Figure 1.4). This is additional to the energy released when both the boiling water and the condensed steam cool on the surface of the skin.
- 11 B
- 12 AWW 01_EX_12

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R Bogdan



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Problems and Problem Solving in Chemistry Education Georgios Tsaparlis, 2021-05-17 Problem solving is central to the teaching and learning of chemistry at secondary tertiary and post tertiary levels of education opening to students and professional chemists alike a whole new world for analysing data looking for patterns and making deductions As an important higher order thinking skill problem solving also constitutes a major research field in science education Relevant education research is an ongoing process with recent developments occurring not only in the area of quantitative computational problems but also in qualitative problem solving The following situations are considered some general others with a focus on specific areas of chemistry quantitative problems qualitative reasoning metacognition and resource activation deconstructing the problem solving process an overview of the working memory hypothesis reasoning with the electron pushing formalism scaffolding organic synthesis skills spectroscopy for structural characterization in organic chemistry enzyme kinetics problem solving in the academic chemistry laboratory chemistry problem solving in context team based active learning technology for molecular representations IR spectra simulation and computational quantum chemistry tools The book concludes with methodological and epistemological issues in problem solving research and other perspectives in problem solving in chemistry With a foreword by George Bodner **The Pearson Guide To The Scra Examination, 2/E**

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Dinesh, 2007-02 The second edition of The Pearson Guide to Complete Mathematics for AIEEE retains the basic structure and coverage of the previous edition while adding to it solved question papers of AIEEE 2005 and 2006 Spread over thirty two systematic and well written chapters this book covers the AIEEE syllabus completely and will also prove a useful guide for students appearing for state level engineering tests PETs *Master the PCAT* Peterson's, 2012-07-15 Peterson's Master the PCAT is an in depth review that offers thorough preparation for the computer based exam After learning about the structure format scoring and score reporting and the subtests and question types you can take a diagnostic test to learn about your strengths and weaknesses The next six parts of the eBook are focused on detailed subject reviews for each subtest verbal ability reading comprehension biology chemistry quantitative ability and writing Each review includes practice questions with detailed answer explanations You can take two practice tests to track your study progress The tests also offer detailed answer explanations to further improve your knowledge and understanding of the tested subjects The eBook concludes with an appendix that provides helpful information on a variety of careers in pharmacy and ten in depth career profiles **The Challenges of Content Acquisition in a CLIL Course** Beata Nawrot-Lis, 2019-10-17 The book deals with content

acquisition in a Content and Language Integrated Learning CLIL based chemistry course and represents high level scientific work The topic of the thesis is current and relevant in the context of up to date research in theory of language education The main research problem stated and discussed in the thesis confirms that all the attempts of introducing CLIL should be valued therefore the book may inspire foreign language teachers and contribute to better understanding of CLIL nature

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 - The Rise of Digital Reading Pearson Chemistry Assessment Answers Problems
 - Advantages of eBooks Over Traditional Books
2. Identifying Pearson Chemistry Assessment Answers Problems
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Pearson Chemistry Assessment Answers Problems
 - User-Friendly Interface
4. Exploring eBook Recommendations from Pearson Chemistry Assessment Answers Problems
 - Personalized Recommendations
 - Pearson Chemistry Assessment Answers Problems User Reviews and Ratings
 - Pearson Chemistry Assessment Answers Problems and Bestseller Lists
5. Accessing Pearson Chemistry Assessment Answers Problems Free and Paid eBooks

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 - Pearson Chemistry Assessment Answers Problems Compatibility with Devices
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- 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pearson Chemistry Assessment Answers Problems
 - Highlighting and Note-Taking Pearson Chemistry Assessment Answers Problems
 - Interactive Elements Pearson Chemistry Assessment Answers Problems
- 8. Staying Engaged with Pearson Chemistry Assessment Answers Problems
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pearson Chemistry Assessment Answers Problems
- 9. Balancing eBooks and Physical Books Pearson Chemistry Assessment Answers Problems
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pearson Chemistry Assessment Answers Problems
- 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
- 11. Cultivating a Reading Routine Pearson Chemistry Assessment Answers Problems
 - Setting Reading Goals Pearson Chemistry Assessment Answers Problems
 - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Pearson Chemistry Assessment Answers Problems
 - Fact-Checking eBook Content of Pearson Chemistry Assessment Answers Problems
 - Distinguishing Credible Sources
- 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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