

Name: KEY

15 Pts.

Chemistry Practice: Writing Chemical Formulas

Write a chemical formula for each substance.

1. NaCl sodium chloride
2. N_2O_5 dinitrogen pentoxide
3. H_2S hydrosulfuric acid
4. K_2SO_4 potassium sulfate
5. $\text{H}_2\text{C}_2\text{O}_4$ oxalic acid
6. $\text{Ag}_2\text{C}_2\text{H}_3\text{O}_2$ silver acetate
7. $\text{Cr}(\text{NO}_3)_3$ chromium(III) nitrate
8. $\text{HC}_2\text{H}_3\text{O}_2$ acetic acid
9. $(\text{NH}_4)_2\text{CO}_3$ ammonium carbonate
10. $\text{Ca}(\text{OH})_2$ calcium hydroxide
11. $\text{H}_2\text{C}_4\text{H}_4\text{O}_6$ tartaric acid
12. $\text{Hg}(\text{NO}_3)_2$ mercury(II) nitrate
13. N_2O dinitrogen monoxide
14. Fe_2O_3 iron(III) oxide
15. $\text{Pb}(\text{ClO}_2)_2$ lead(II) chlorate
16. $(\text{NH}_4)_3\text{PO}_4$ ammonium phosphate
17. ZnCl_2 zinc chloride
18. $\text{Ca}_3(\text{PO}_4)_2$ calcium phosphate
19. OF_2 oxygen difluoride
20. NaN_3 sodium azide
21. $\text{Fe}_2(\text{SO}_4)_3$ iron(III) sulfate
22. H_3AsO_3 arsenous acid
23. Cr_2O_3 chromium(III) oxide
24. N_2O_4 dinitrogen tetroxide
25. NH_4NO_3 ammonium nitrate
26. AuBr_3 gold(III) bromide
27. CO carbon monoxide
28. K_2CO_3 potassium carbonate
29. HIO_3 iodic acid
30. CsCl cesium chloride
31. $\text{Ni}(\text{MnO}_4)_2$ nickel(II) permanganate
32. $\text{Al}_2(\text{SO}_4)_3$ aluminum sulfate
33. $\text{Al}_2(\text{SO}_3)_3$ aluminum sulfite
34. $\text{Ba}(\text{C}_2\text{H}_3\text{O}_2)_2$ barium acetate
35. $\text{Mn}(\text{OH})_2$ manganese(II) hydroxide
36. KH_2PO_4 potassium dihydrogen phosphate
37. HF hydrofluoric acid
38. $\text{Al}(\text{BrO}_3)_3$ aluminum bromate

39. $\text{Ca}(\text{NO}_3)_2$ calcium nitrate
40. SO_3 sulfur trioxide
41. KCN potassium cyanide
42. $\text{Pb}(\text{NO}_3)_2$ lead(II) nitrate
43. H_2S hydrogen sulfide
44. CoCl_3 cobalt(III) chloride
45. SF_6 sulfur hexafluoride
46. Ca_3N_2 calcium nitride
47. CuI copper(I) iodide
48. SiO_2 silicon dioxide
49. $\text{Sn}(\text{C}_2\text{H}_3\text{O}_2)_4$ tin(IV) acetate
50. CCl_4 carbon tetrachloride
51. CuS copper(II) sulfide
52. $\text{Pb}_2(\text{PO}_4)_2$ lead(II) phosphate
53. XeCl_4 xenon tetrachloride
54. Rb_2O rubidium oxide
55. MgSe magnesium selenide
56. NH_4Cl ammonium chloride
57. $\text{Fe}(\text{C}_2\text{H}_3\text{O}_2)_3$ iron(III) acetate
58. $\text{K}_2\text{Cr}_2\text{O}_7$ potassium dichromate
59. PBr_3 phosphorous tribromide
60. Na_3PO_3 sodium phosphite
61. Na_3PO_4 sodium phosphate
62. $\text{Hg}(\text{NO}_3)_2$ mercury(II) nitrate
63. LiHCO_3 lithium bicarbonate
64. CrF_3 chromium(III) fluoride
65. PbI_2 lead(II) iodide
66. H_2SO_3 sulfurous acid
67. SnF_2 tin(II) fluoride
68. HgCrO_4 mercury(II) chromate
69. KNO_3 potassium nitrate
70. SrCl_2 strontium chloride
71. P_4O_{10} tetraphosphorous decoxide
72. KNO_3 potassium nitrate
73. KNO_2 potassium nitrite
74. K_3N potassium nitride
75. CaO calcium oxide
76. $\text{Fe}(\text{IO}_4)_2$ iron(II) periodate

Pearson Education Chemical Names And Formulas Answers

Lorraine Mirza, Krishnakali Gupta



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