

Reviewing Content

42. The solvent is the substance in which the solute is dissolved.
43. Random collisions of the solvent molecules with the solute particles provide enough force to overcome gravity.
44. **solubility**: the amount of a substance that dissolves in a given quantity of solvent at specified conditions of temperature and pressure to produce a saturated solution.
saturated solution: a solution containing the maximum amount of solute for a given amount of solvent at a constant temperature and pressure. **unsaturated solution**: a solution that contains less solute than a saturated solution at a given temperature and pressure. **miscible**: describes liquids that dissolve in each other. **immiscible**: describes liquids that are insoluble in each other.
45. Particles of solute crystallize.
46. No; if there were undissolved solute, the excess solute would come out of a supersaturated solution.
47. 5.55×10^2 g AgNO_3
48. Solubility increases with pressure.
49. a. 1.6×10^{-2} g/L
b. 4.7×10^{-2} g/L
50. *Dilute* and *concentrated* are relative terms and are not quantitative. Molarity provides the exact number of moles of solute per liter of solution.
51. Molarity is the number of moles of solute dissolved in one liter of solution.
a. 1.3M KCl
b. 3.3×10^{-1} M MgCl_2
52. 2.00×10^3 mL
53. a. 5.0×10^{-3} mol NaCl, 29 g NaCl
b. 1.0 mol KNO_3 , 1.0×10^{-2} g KNO_3
c. 2.5×10^3 mol CaCl_2 , 2.8 g CaCl_2
54. a. 2.3×10^1 g NaCl
b. 2.0 g MgCl_2
55. a. 16% (v/v) ethanol
b. 63.6% (v/v) isopropyl alcohol
56. Colligative properties are properties of a solution that depend only on the number of solute particles; boiling-point elevation, freezing-point depression, and vapor-pressure lowering. Boiling points are elevated because shells of solvent form around solute particles, reducing the amount of solvent molecules that have sufficient energy to escape the solution; relative to the pure solvent, the amount of energy required to cause vaporization or boiling increases. Solutes disrupt the ordering of the solvent structure, so more kinetic energy must be withdrawn from a solution for it to solidify. This lowers the freezing point of the solution.
57. a. sea water
b. 1.5M KNO_3
c. 0.100M MgCl_2
58. The effective molality of the $\text{Ca}(\text{NO}_3)_2$ solution is 3m. The effective molality of the NaNO_3 solution is 2m.
59. When vapor pressure is lowered relative to pure solvent, more energy must be supplied to reach the boiling point; thus the boiling point is increased relative to pure solvent.
60. The salt lowers the freezing point of the ice-water cooling mixture.
61. 1M solution: 1 mol of solute in 1 L of solution; 1m solution: 1 mol of solute in 1000 g of solvent
62. Add 27.0 g H_2O to 32.0 g CH_3OH .
63. a. 100.26°C
b. 101.54°C
64. a. -4.46°C
b. -2.2°C
65. a. -1.1°C
b. -0.74°C
c. -1.5°C

Understanding Concepts

66. a. The freezing-point depression is twice as great for solute B; solute B must provide twice as many particles in solution.

Pearson Education Cheistry Reading Guide Answers

Terry C. Jones



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Study Guide with Selected Solutions Karen Timberlake, 2002-08-06 Written by the author the Study Guide is keyed to the learning goals in the text and designed to promote active learning through a variety of exercises with answers and mastery exams Also contains complete solutions to odd numbered problems *General, Organic, and Biological Chemistry Study Guide and Selected Solutions* Karen C. Timberlake, 2001-11 Keyed to the learning goals in the text this guide is designed to promote active learning through a variety of exercises with answers and mastery exams The guide also contains complete solutions to odd numbered problems Handbook of Reading Research Michael L. Kamil, Peter B. Mosenthal, P. David Pearson, Rebecca Barr, 2014-04-08 In Volume III as in Volumes I and II the classic topics of reading are included from vocabulary and comprehension to reading instruction in the classroom and in addition each contributor was asked to include a brief history that chronicles the legacies within each of the volume s many topics However on the whole Volume III is not about tradition Rather it explores the verges of reading research between the time Volume II was published in 1991 and the research conducted after this date The editors identified two broad themes as representing the myriad of verges that have emerged since Volumes I and II were published 1 broadening the definition of reading and 2 broadening the reading research program The particulars of these new themes and topics are addressed **Loyola Educational Index**, 1928 Resources in Education, 1999 **Focus on Physical Science California Edition** Michael J. Padilla, 2007 Reader's Guide to Periodical Literature Supplement, 1926 **The Pearson Guide To The Scra Examination, 2/E** Thorpe, 2008-09 **The Burning Answer** Keith Barnham, 2014-05-15 Our civilisation stands on the brink of catastrophe Our thirst for energy has led to threats from global warming nuclear disaster and conflict in oil rich countries We are running out of options Solar power Keith Barnham argues is the answer In this eye opening book he shows how a solar revolution is developing based on one of Einstein s lesser known discoveries one that gave us laptop computers and mobile phones An accessible guide to renewable technology and a hard hitting critique of the arguments of solar sceptics The Burning Answer outlines a future in which the fuel for electric cars will be generated on our rooftops It is above all an impassioned call to arms to join the solar revolution before it s too late **The Pearson Guide to MCA Entrance Examinations** Thorpe, *Catalog of Copyright Entries. Third Series* Library of Congress. Copyright Office, 1964 Includes Part 1 Number 1 Books and Pamphlets Including Serials and Contributions to Periodicals January June **Working with Text and Around Text in Foreign Language Environments** Halina Chodkiewicz, Piotr Steinbrich, Małgorzata Krzemińska-Adamek, 2016-06-28 This book investigates the three pivotal points of text for foreign language acquisition reception construction and deconstruction In Part One the focus is on various aspects of text reception such as developing literacy text interest and perceptions of the academic register or the assessment of spoken language in educational contexts Part Two deals with various aspects of composing text such as author identity lexical constructs or collaborative web based writing Lastly Part Three presents the various segmental items

that constitute text like lexical clustering L1 L2 relationship classroom talk as text etc The division corresponds with what can be viewed as a logical sequence of text related processes reflected in formal learning and teaching environments

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