

Chapter 1

CHEMICAL REACTIONS AND EQUATIONS

Objective Questions

Multiple Choice Questions

- Which of the statements about the reaction below are incorrect?
$$2\text{PbO(s)} + \text{C(s)} \longrightarrow 2\text{Pb(l)} + \text{CO}_2\text{(g)}$$
 - (i) Lead is getting reduced
 - (ii) Carbon dioxide is getting oxidised
 - (iii) Carbon is getting oxidised
 - (iv) Lead oxide is getting reduced

(a) (i) and (ii) (b) (i) and (iii)
(c) (i), (ii) and (iii) (d) all of these.
- $$\text{Fe}_2\text{O}_3 + 2\text{Al} \longrightarrow \text{Al}_2\text{O}_3 + 2\text{Fe}$$

The above reaction is an example of a— (2024)

 - (a) combination reaction
 - (b) double displacement reaction
 - (c) decomposition reaction
 - (d) displacement reaction.
- What happens when dilute hydrochloric acid is added to iron filings? (2022)

 - (a) Hydrogen gas and iron chloride are produced
 - (b) Chlorine gas and iron hydroxide are produced
 - (c) No reaction takes place
 - (d) Iron salt and water are produced.

- In the reaction $2\text{FeCl}_2 + \text{Cl}_2 \longrightarrow 2\text{FeCl}_3$ Chlorine is a—

 - (a) Reducing agent
 - (b) Oxidizing agent
 - (c) Catalyst
 - (d) Inert medium.

- Rancidity can be prevented in food items by—

 - (a) Mixing antioxidant
 - (b) Keeping in refrigerator
 - (c) Keeping away from light
 - (d) All of these.

- Which among the following is not an oxidizing agent?

 - (a) Oxygen
 - (b) Conc. Sulphuric acid
 - (c) Chlorine
 - (d) Hydrogen.

- Which among the following gases are used to keep the oily food items fresh?

 - (a) Carbon dioxide and Oxygen
 - (b) Nitrogen and helium
 - (c) Helium and Oxygen
 - (d) Nitrogen and Oxygen.

- Chemically rust is—

 - (a) Hydrated ferrous oxide
 - (b) Ferric oxide
 - (c) Hydrated ferric oxide
 - (d) None of these.

- An exothermic reaction in which glucose combines with oxygen in the cells of our body and provides energy—

 - (a) Respiration
 - (b) Digestion
 - (c) Excretion
 - (d) None of these.

Pearson Education Chapter 1 chemical Reactions Key

Michael Brown



Pearson Education Chapter 1chemical Reactions Key:

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