

List of Common Polyatomic Ions

Cations:

<u>Symbol</u>	<u>Name</u>	<u>Charge</u>	<u>Symbol</u>	<u>Name</u>	<u>Charge</u>
NH_4^+	Ammonium	+1	Hg_2^{+2}	Dimercury (I)	+2

Anions:

<u>Symbol</u>	<u>Name</u>	<u>Charge</u>	<u>Symbol</u>	<u>Name</u>	<u>Charge</u>
$\text{C}_2\text{H}_3\text{O}_2^-$	Acetate	-1	CO_3^{-2}	Carbonate	-2
HCO_3^-	Bicarbonate	-1	CrO_4^{-2}	Chromate	-2
BrO_3^-	Bromate	-1	$\text{Cr}_2\text{O}_7^{-2}$	Dichromate	-2
BrO_2^-	Bromite	-1	$\text{C}_2\text{O}_4^{-2}$	Oxalate	-2
ClO_4^-	Perchlorate	-1	O_2^{-2}	Peroxide	-2
ClO_3^-	Chlorate	-1	SO_4^{-2}	Sulfate	-2
ClO_2^-	Chlorite	-1	SO_3^{-2}	Sulfite	-2
ClO^-	Hypochlorite	-1	$\text{S}_2\text{O}_3^{-2}$	Thiosulfate	-2
CN^-	Cyanide	-1	SiO_3^{-2}	Silicate	-2
SCN^-	Thiocyanate	-1	MoO_4^{-2}	Molybdate	-2
OH^-	Hydroxide	-1	WO_4^{-2}	Tungstate	-2
NO_3^-	Nitrate	-1	BO_3^{-3}	Borate	-3
NO_2^-	Nitrite	-1	PO_4^{-3}	Phosphate	-3
MnO_4^-	Permanganate	-1	PO_3^{-3}	Phosphite	-3
HSO_4^-	Bisulfate	-1	AsO_4^{-3}	Arsenate	-3
HSO_3^-	Bisulfite	-1	AsO_3^{-3}	Arsenite	-3
IO_3^-	Iodate	-1			
IO_4^-	Periodate	-1			
$\text{C}_6\text{H}_5\text{COO}^-$	Benzoate	-1			

<u>Symbol</u>	<u>Name</u>	<u>Charge</u>
$\text{H}_2\text{PO}_4^{-1}$	Dihydrogen phosphate	-1
HPO_4^{-2}	Hydrogen phosphate	-2
$\text{Fe}(\text{CN})_6^{-3}$	Hexacyanoferrate (III)	-3
$\text{Fe}(\text{CN})_6^{-4}$	Hexacyanoferrate (II)	-4

Polyatomic Ions Packet Answers

Sally M. McIntyre

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

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