

Chemistry Worksheets Class 12 on Chapter 15 Polymers - Set 1

Q-1: $F_2C=CF_2$ is a monomeric unit of ____.

- a.) Teflon
- b.) Nylon-6
- c.) Glyptal
- d.) Buna-S

Q-2: The polymer used in the bullet-proof glasses is:

- a.) PMMA
- b.) Lexan
- c.) Kevlar
- d.) Nomex

Q-3: What is the difference between Buna-N and Buna-S?

Q-4: Nylon-6 is prepared from:

- a.) Chloroprene
- b.) Adipic acid
- c.) 1,3-Butadiene
- d.) Caprolactam

Q-5: How can the functionality of a monomer be explained?

Q-6: Which of the following polymers is not a thermosetting polymer?

- a.) PVC
- b.) Bakelite
- c.) Epoxy resin
- d.) Melmac

Q-7: What is the common name and the IUPAC name of the monomer of natural rubber?

Q-8: What is the role of sulphur in vulcanization of rubber?

Q-9: Chloroprene is a repeating unit in ____.

Polymer Chemistry Questions And Manual Answers

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Table of Contents Polymer Chemistry Questions And Manual Answers

1. Understanding the eBook Polymer Chemistry Questions And Manual Answers
 - The Rise of Digital Reading Polymer Chemistry Questions And Manual Answers
 - Advantages of eBooks Over Traditional Books
2. Identifying Polymer Chemistry Questions And Manual Answers
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an eBook Platform
 - User-Friendly Interface
4. Exploring eBook Recommendations from Polymer Chemistry Questions And Manual Answers
 - Personalized Recommendations
 - Polymer Chemistry Questions And Manual Answers User Reviews and Ratings
 - Polymer Chemistry Questions And Manual Answers and Bestseller Lists

5. Accessing Polymer Chemistry Questions And Manual Answers Free and Paid eBooks
 - Polymer Chemistry Questions And Manual Answers Public Domain eBooks
 - Polymer Chemistry Questions And Manual Answers eBook Subscription Services
 - Polymer Chemistry Questions And Manual Answers Budget-Friendly Options
6. Navigating Polymer Chemistry Questions And Manual Answers eBook Formats
 - ePub, PDF, MOBI, and More
 - Polymer Chemistry Questions And Manual Answers Compatibility with Devices
 - Polymer Chemistry Questions And Manual Answers Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Polymer Chemistry Questions And Manual Answers
 - Highlighting and Note-Taking Polymer Chemistry Questions And Manual Answers
 - Interactive Elements Polymer Chemistry Questions And Manual Answers
8. Staying Engaged with Polymer Chemistry Questions And Manual Answers
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Polymer Chemistry Questions And Manual Answers
9. Balancing eBooks and Physical Books Polymer Chemistry Questions And Manual Answers
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Polymer Chemistry Questions And Manual Answers
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Polymer Chemistry Questions And Manual Answers
 - Setting Reading Goals Polymer Chemistry Questions And Manual Answers
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Polymer Chemistry Questions And Manual Answers
 - Fact-Checking eBook Content of Polymer Chemistry Questions And Manual Answers
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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 $2 \text{ CO}_2(\text{g}) + \text{H}_2\text{O}(\text{g}) \rightarrow \text{C}_2\text{H}_2(\text{g}) + \frac{5}{2} \text{O}_2(\text{g})$ $\Delta H = 94.5 \text{ kJ}$
 $\text{H}_2\text{O}(\text{g}) \rightarrow \text{H}_2(\text{g}) + \frac{1}{2} \text{O}_2(\text{g})$ $\Delta H = 71.2 \text{ kJ}$
 $\text{C}_2\text{H}_6(\text{g}) + \frac{7}{2} \text{O}_2(\text{g}) \rightarrow \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{g})$ $\Delta H = 283 \text{ kJ}$

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 $\text{C}_2\text{H}_4(\text{g}) + 3 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 2 \text{H}_2\text{O}(\text{l})$ $\Delta H = 1411 \text{ kJ/mole}$
 $\text{C}_2\text{H}_6(\text{g}) + 7/2 \text{O}_2(\text{g}) \rightarrow 2 \text{CO}_2(\text{g}) + 3 \text{H}_2\text{O}(\text{l})$ $\Delta H = 1560 \text{ kJ/mole}$
 $\text{H}_2(\text{g}) + 1/2 \text{O}_2(\text{g}) \rightarrow \text{H}_2\text{O}(\text{l})$ $\Delta H = 285 \text{ kJ/mole}$

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