

Chapter 12 – Stoichiometry

Note Taking Guide: Episode 801 Stoichiometry

-Study of the **QUANTITY** relationships in a **CHEMICAL REACTION**

-Based on **BALANCED** equations



1. Particles

2 atoms Mg	2 atoms O	2 atoms Mg
		2 atoms O

2 atoms Mg	1 molecule O ₂	2 molecule MgO
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10 atoms Mg	5 molecule O ₂	10 molecule MgO
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2 (6.02 x 10 ²³) atoms Mg	1 (6.02 x 10 ²³) molecule O ₂	2 (6.02 x 10 ²³) molecule MgO
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2. Moles

2 mol Mg	1 mol O ₂	2 mol MgO
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The **COEFFICIENTS** in a **BALANCED EQUATION** give the **MOLE RATIO** for the **SUBSTANCES** involved in the **REACTION**.

Lab: Episode 801 - Can You Find the Mole Ratio?

In this experiment, baking soda (NaHCO₃) reacts with hydrochloric acid (HCl) to produce sodium chloride, carbon dioxide, and water. You will determine the moles of reactant used and moles of product produced using both the lab data and stoichiometry and compare the two.

Data Table:

mass of test tube and baking soda	19.46 g
mass of empty test tube	19.21 g

Note Taking Guide Episode 801 Stoichiometry

A Loxley



Note Taking Guide Episode 801 Stoichiometry:

Unveiling the Magic of Words: A Overview of "**Note Taking Guide Episode 801 Stoichiometry**"

In some sort of defined by information and interconnectivity, the enchanting power of words has acquired unparalleled significance. Their power to kindle emotions, provoke contemplation, and ignite transformative change is actually awe-inspiring. Enter the realm of "**Note Taking Guide Episode 801 Stoichiometry**," a mesmerizing literary masterpiece penned by way of a distinguished author, guiding readers on a profound journey to unravel the secrets and potential hidden within every word. In this critique, we shall delve in to the book is central themes, examine its distinctive writing style, and assess its profound impact on the souls of its readers.

<https://crm.avenza.com/files/detail/default.aspx/phenom%20pilot%20manual.pdf>

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