

Chemical Formula

Talking about proportions, molecules and relative all atoms present (in a compound)

Molecular $\text{C}_{10}\text{H}_{12}\text{O}_4$ (molecular)

Formula represents the number of atoms of each element in a single molecule

Empirical $\text{C}_5\text{H}_6\text{O}_2$ (molecular)

Formula represents the simplest ratio of atoms of each element present and negative charge formula

Atomic Mass

The average mass of atoms present together

Mass

$\text{C}_{10}\text{H}_{12}\text{O}_4$ mass

Mass of atoms present

Substance

Mass of atoms

Mass of atoms present

Mass

Note Taking Guide Episode 601 Answers Physics

SB Merriam



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Unveiling the Power of Verbal Artistry: An Mental Sojourn through **Note Taking Guide Episode 601 Answers Physics**

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